## Propane

## Performance:

This camping-type stove is low powered. In Mexico, some propane stoves are not hot enough to make tortillas, so the 3 Stone Fire is used.

Cooking on a propane stove is quite luxurious after operating a woodburning stove. It is pleasurable to turn on the stove and cook food without having to even think about tending the fire.


Propane

Propane can be somewhat dangerous as old storage cylinders and stoves begin to leak.

Time to boil is slightly faster than the 3 Stone Fire. Emissions are close to zero compared to the other stoves in these tests.

## Test Results



## Alcohol - Clean Cook Prototype

Origin: Nigeria
Weight: 5.1 kilos
Fuel Type: Alcohol
Contact:
Project Gaia, Inc.
Mr. Harry Stokes
22 Mummasburg Street,
PO Box 4190
Gettysburg, PA 17325
hstokes@projectgaia.com www.projectgaia.com tel: (717) 334-5594
fax: (717) 334-7313


## Description:

The Clean Cook alcohol stove prototype has two large fuel tanks filled with an absorptive material so the filled tank can be placed under the burner without leaking. Protective barriers placed over the fuel canisters prevent filling the stove while lit.

The stove body is stainless steel and attractively made.Two levers open and close the burners. By adjusting the levers, which close a cover over the fire, the power can be controlled. Two sheet metal pot supports help to shield the fire.

The tanks are not pressurized, allowing the fuel to burn in small, open cylinders underneath the pots. An unpressurized system is simple and does not depend on air tightness to work.

## Alcohol - Clean Cook Prototype

## Performance:

Alcohol has been a popular fuel for many years. Like kerosene, it has been used on boats when propane is considered too dangerous.

Alcohol stoves have a reputation for being somewhat low powered. In this case, the pot used in the Water Boiling Test was covered, which helped


Alcohol-clean cook the water reach full boil but makes comparisons with other stoves using uncovered pots problematic. The lid was removed for simmering.

The stove cooks food like other liquid-fueled stoves, without tending. The cook can work on other tasks and gain hours once spent adjusting the fire.

## Test Results



## Emissions

Carbon Monoxide (CO)


Particulate Matter (PM)


## Kerosene

Origin: Hong Kong
Weight: 2.3 kilos
Fuel Type: Kerosene
Contact:
Solar Stoves
B/32 Shanker Tekari Jamnagar, Gujarat
India 361004


## Description:

This kerosene stove uses wicks to bring kerosene into a combustion chamber where, with proper adjustments such as trimming the wicks and having the wicks at the appropriate height, a blue flame is created under the pot.

When correctly adjusted, the stove can burn cleanly. However, as received, the stove is somewhat smoky.

An adjustable knob moves the multiple wicks up and down. In this way, higher and lower power can be achieved. The wicks release the correct amount of gases that combust in a vertical chimney. The evolution of this simple-but-effective system has created a remarkable technology that effectively burns the unpressurized fuel.

A large tank under the stove holds the kerosene. The stove body is made of painted sheet metal.

## Kerosene

## Performance:

The short internal chimney helps the kerosene stove burn with a blue flame when adjusted correctly. To operate the stove cleanly may take time and practice, and the stove may need to be rebuilt. As received, the stove was smoky and the stove tank leaked.

However, after the stove was set up properly, it ran well without much tending.


Emissions, while low, are appreciably higher than with propane and alcohol.

## Test Results

## Fuel Economy

## (D) Time to Boil 5L of Water - 41:54

Energy Consumption compared to other stoves


Fuel Used to Boil
115 g

+ Fuel Used to Simmer
132 g
$5 L$ of Water for 45 minutes
TOTAL -
247 g
Cost $=\mathbf{\$ 9 . 5 0}$
Fuel Use = $\mathbf{1 5} \mathbf{~ k g}$
estimated per month
Safety Rating



## Emissions



48\%

Particulate Matter (PM)


20\%

Parabolic Solar Cooker<br>Origin: USA<br>Weight: 103 kilos<br>Fuel Type: Sunshine<br>Contact:<br>Aprovecho Research Center<br>PO Box 1175<br>Cottage Grove, Oregon 97424<br>www.aprovecho.org<br>tel: 541 767-0287



## Description:

The parabolic solar cooker was built by students at Aprovecho. It was made from a recycled fiberglass satellite dish six feet in diameter. One-inch by one-inch square mirrors were glued to the surface with silicon adhesive.

A metal support holds an insulated box with a glass bottom at the focal point. Reflected sunlight passes through glass on the bottom of the insulated box. The insulation helps capture the heat and increase the efficiency of heat transfer. The box is placed around the pot and can be removed by the cook.

The parabola is supported inside a wooden frame on rollers so the reflector can follow the moving sun. The solar cooker needs to be re-aimed at the sun about every half hour. The stove can also be used with a wok for grilling. One or two pounds of food can be fried quite successfully.

## Parabolic Solar Cooker

## Performance:

The solar cooker can generate over 2,000 watts of power, boiling 5 L of water in an average of 70 minutes.

Solar cooking uses no fuel and makes no emissions. The solar cooker is the only stove tested that does not use diminishing resources to cook food. The fuel is free, as long as the sun is shining.


The cook can usually stand behind the reflector to stir food. However, the glare when standing in front of the dish can be intense.

It is necessary to be careful, because the heat at the focal point is invisible and over $550^{\circ} \mathrm{C}$.
Cooking with this parabolic dish is easy compared to using wood, because it only requires tracking the sun once in a while.

## Test Results



## Chapter 2

## Stove Rankings

This chapter contains lists and graphs showing how each stove ranks in eight important performance categories:

## 1. Time to Boil

Waiting for a pot to boil, or for tortillas to cook on a slow stove, can be frustrating. Cooks and families often appreciate a powerful, adjustable stove.

## 2. Fuel to Cook and 3. Energy to Cook

When looking at fuel consumption, it is important to consider the amount of energy in each type of fuel. For instance, propane has over twice the amount of useful energy in each gram compared to wood. When a particular stove uses less fuel, it does not necessarily use less energy.

## 4. Carbon Monoxide and 5. Particulate Matter Emissions

Carbon monoxide (CO) is a deadly odorless, poisonous gas. Inhaling particulate matter (PM) can cause acute respiratory infections and a host of other diseases. To protect the health of a family, high levels of indoor air pollution must be prevented. Please note that measures of particulate matter include total emissions produced by the stoves, even chimney stoves, which protect the user from these emissions. For this reason, while the chimney took almost all of the pollution out of the test kitchen, the PM results are higher, as measured under the collection hood (see page 83) from the chimney exit.

## 6. Safety Ratings

Using fire can be dangerous. Burns are often horribly disfiguring. A stove should be as safe as possible. Stoves were evaluated for safety using safety evaluation methods developed by Nathan Johnson at Iowa State University. Details on the evaluation procedures can be found in Appendix C on page 121 .

## 7. Cost to Purchase and 8. Monthly Fuel Use

The cost to build or purchase a stove and the continual burden of fuel costs can be very important factors in stove choice. A more expensive stove that saves money by using less fuel can be a worthwhile purchase. However, if the initial cost is too high, the stove may never become popular. If fuel is scarce in the area where the stove is being used, fuel use may be the most important factor.

## What is the best stove for you?

Some people may think that the cost of the stove is most important. Others might put a higher value on time to boil, fuel use or safety. Which categories are most important to your market?

A total value for each stove can be determined by adding the score in each of the categories that are most important to you. The best stove in each category can be given a score of 1 , the next 2 , and so on. In this way, the stove with the lowest total score would be the "best" and might suit your needs. As we've said, the choice of a best stove may be based on preferences that are outside of these categories. For example, griddle stoves can make tortillas and simmer multiple pots using one fire. The griddle stove uses more energy to boil a single pot, but it may cook food more successfully.

We strongly recommend that local cooks try the proposed stove. Only cooks will know if a stove is suitable or not.

## 1. Time to Boil

## 5 L of water

## Shortest Time

1. Vita Stove
2. Mud/Sawdust
3. Uganda 2-pot
4. Wood Flame Fan
5. Ghana Wood
6. 20 L Can Rocket
7. Propane
8. Wood Gas Fan
9. 3 Stone Fire
10. Onil
11. Gyapa Charcoal
12. Alcohol - Clean Cook
13. Patsari Prototype
14. Mali Charcoal
15. Ecostove
16. Kerosene
17. Justa
18. Parabolic Solar Cooker

Longest Time



## 2. Fuel to Cook

## Boil and Simmer 5 L of water for 45 minutes

## Least Fuel Used

1. Parabolic Solar Cooker
2. Propane
3. Kerosene
4. Alcohol - Clean Cook
5. Wood Gas Fan
6. Wood Flame Fan
7. Mali Charcoal
8. VITA
9. Gyapa Charcoal
10. Uganda 2-pot
11. 20 L Can Rocket
12. Mud/Sawdust
13. Ghana Wood
14. 3 Stone Fire
15. Patsari Prototype
16. Justa
17. Onil
18. Ecostove
\#1 Parabolic Solar Cooker 0.0 grams



## 3. Energy to Cook <br> Boil and Simmer 5 L of water for 45 minutes

## Least Energy

1. Parabolic Solar Cooker
2. Propane
3. Alcohol - Clean Cook
4. Wood Gas Fan
5. Kerosene
6. Wood Flame Fan
7. Uganda 2-pot
8. VITA
9. 20 L Can Rocket
10. Mud/Sawdust
11. Ghana Wood
12. Gyapa Charcoal
13. 3 Stone Fire
14. Mali Charcoal
15. Patsari Prototype
16. Justa
17. Onil
18. Ecostove

\#3 Alcohol Clean Cook 6,766 kJ


## Most Energy



## 4. Carbon Monoxide Emissions Boil and Simmer 5 L of water for 45 minutes

## Least CO Released

1. Parabolic Solar Cooker
2. Propane
3. Alcohol - Clean Cook
4. Wood Gas Fan
5. Kerosene
6. Wood Flame Fan
7. 20 L Can Rocket
8. Patsari Prototype
9. Uganda 2-pot
10. Justa
11. Onil
12. VITA
13. Ecostove
14. Mud/Sawdust
15. Ghana Wood
16. 3 Stone Fire
17. Mali Charcoal
18. Gyapa Charcoal

Most CO Released

## \#1 Parabolic Solar Cooker 0.0 grams


\#2 Propane
0.5 grams


## \#3 Alcohol Clean Cook 5.3 grams



Please note that these are measures of total emissions produced by the stove (including emissions that would normally be exhausted from the house via a chimney), not emissions to which the cook is exposed. Many chimney stoves that resulted in low emissions in the test kitchen measure higher in the PM and CO emissions categories, which were measured under the collection hood from the chimney exit.


## 5. Particulate Matter Emissions

## Boil and Simmer 5 L of water for 45 minutes

## Least PM Released

1. Parabolic Solar Cooker
2. Alcohol - Clean Cook
3. Propane
4. Kerosene
5. Wood Gas Fan
6. Wood Flame Fan
7. Mali Charcoal
8. Gyapa Charcoal
9. Uganda 2-pot
10. Justa
11. Patsari
12. 20 L Can Rocket
13. Onil
14. VITA
15. Mud/Sawdust
16. 3 Stone Fire
17. Ghana Wood
18. Ecostove

Most PM Released

## \#1 Parabolic Solar Cooker 0.0 mg



Please note that these are measures of total emissions produced by the stove (including emissions that would normally be exhausted from the house via a chimney), not emissions to which the cook is exposed. Many chimney stoves that resulted in low emissions in the test kitchen measure higher in the PM and CO emissions categories, which were measured under the collection hood from the chimney exit.


## 6. Safety Ratings <br> Evaluated on 10 criteria (see Appendix)

## Most Safe

1. Onil
2. Justa
3. Alcohol - Clean Cook
4. Uganda 2-pot
5. Patsari Prototype
6. Kerosene
7. Wood Flame Fan
8. Ecostove
9. Propane
10. Mali Charcoal
11. Wood Gas Fan
12. Mud/Sawdust
13. 20 L Can Rocket
14. Parabolic Solar Cooker
15. Gyapa Charcoal
16. Ghana Wood
17. VITA
18. 3 Stone Fire

Least Safe



## 7. Cost to Purchase <br> In US \$

## Least Expensive

1. 3 Stone Fire
2. Mud/Sawdust
3. 20 L Can Rocket
4. VITA
5. Mali Charcoal
6. Ghana Wood
7. Gyapa Charcoal
8. Kerosene
9. Propane
10. Alcohol - Clean Cook
11. Patsari Prototype
12. Uganda 2-pot
13. Solar Cooker
14. Ecostove
15. Onil
16. Justa
17. Wood Gas Fan
18. Wood Flame Fan


Most Expensive


## 8. Monthly Fuel Use <br> Wood Burning Stoves (kg / month)

## Least Fuel Used

1. Wood Gas Fan
2. Wood Flame Fan
3. VITA
4. Uganda 2-pot
5. 20 L Can Rocket
6. Mud/Sawdust
7. Ghana Wood
8. 3 Stone Fire
9. Patsari Prototype
10. Justa
11. Onil
12. Ecostove

Most Fuel Used



## Chapter 3

## Learning From Improved Cook Stoves

## Why do some wood-burning cook stoves boil water faster?

The 3 Stone Fire (Figure 1) is often thought of as a fast way to boil water. If an improved stove doesn't boil water as quickly, people may switch back to the 3 Stone Fire when they are in a hurry.

An improved stove designed to boil water quickly must have sufficient firepower. The heat created in the stove has to be high enough to cook local foods in


Figure 1 acceptable times. To boil water quickly, as much heat as possible has to get from the fire into the pot. It is important to make sure the flame and hot gases are directed right at the pot. Increasing the temperature of the hot gases helps the stove boil water faster than the 3 Stone Fire.

Eight stoves in these tests boiled 5 L of water faster than the 3 Stone Fire (Figure 2). The three stoves that boiled water the fastest in these tests were the VITA, Mud/Sawdust, and Uganda 2-pot (Figure 3). They each have similar narrow channels around the pot that force the hot gases to flow against the bottom and sides of the pot.

If the channel around the pot is not narrow enough, the hot gases will flow up the middle of the channel, avoiding the surface of the pot. At the same time, it is very important that the increased friction in the narrow channel does not slow the flow of gases and air through the stove too much, otherwise the heat transfer to the pot will be decreased.

The flow of hot gases is like a river of water. The river of gases should not meet a restriction, such as a dam, that would diminish its volume or speed. If the river becomes half as wide, it needs to also become twice as deep to continue flowing at the same speed. In cleaner burning wood-fired stoves, most of the heat is brought to the pot by the hot gases. If the gases move slowly, less heat makes it into the pot.

Gas has very little mass, so the few hot molecules in the moving gases cannot transport much heat energy per volume. It takes a lot of hot gas to deliver the required heat to a pot or griddle. For this reason, more heat is brought to the pot by increasing both the amount and speed of the hot gases without reducing their temperature.

Radiation from the fire can be important in transferring heat, but to be effective, the radiant surface has to be hot and close to the pot. In wood-burning stoves, bringing the pot closer to the fire can increase smoke and harmful pollution. In cleaner burning stoves, the pot is farther away from the fire and is therefore mostly warmed by hot flue gases.

## Four techniques to boil water faster:

1. Create a large enough fire in the combustion chamber.

2. Force the gases to flow against the bottom and sides of the pot in narrow channels.
3. Make sure the gases are as hot as possible.
4. Increase the speed of the hot gases flowing over the surface of the pot.

Figure 2-Stoves that boil 5 L of water faster than the 3 Stone Fire


Figure 3 Fastest to Boil 5 L of Water


## Why do some wood stoves use less fuel?

The 3 Stone Fire (Figure 4) can be fairly fuel efficient when operated carefully. In the Aprovecho laboratory tests, expert operators tried to get optimal results from each stove. The 3 Stone Fire consumed an average of about 1,100 grams of wood to bring to a boil and then simmer 5 L of water for 45 minutes. In the field, the 3 Stone Fire usually consumes more wood.

Six wood-burning stoves in these tests used less fuel to complete the Water Boiling Test. The graph below (Figure 7) details the performance of the wood-burning stoves that used less energy than a 3 Stone Fire. Liquid-fueled stoves and charcoalburning stoves are included to place the results in a wider context.

The Wood Gas (Figure 5) and Wood Flame (Figure 6) stoves use electric fans to improve combustion efficiency. The low-volume, high-velocity jets of air increase the mixing of gas, air and fire in the
combustion chamber. At the same time, the velocity of hot gases contacting the pot is also increased. Even though the hot gases contact only the bottom of the pot, the two stoves consumed the least wood in these tests. Fans seem to tremendously help wood-burning stoves do well in all categories of performance. Adding a fan to a wood-burning stove seems like a great idea from what we have seen in these tests.

The VITA, Uganda 2-pot, 20 L Can Rocket, and Mud/Sawdust are natural-draft stoves. The velocity of the flame and hot gases is determined by the heat of the fire. In these stoves, the heat is forced to contact the sides as well as the bottom of the pot, so more of the heat from the fire gets into the pot. Luckily for stove builders and designers, the four techniques that help a cooking stove boil water faster also help reduce fuel use.


Figure 5


Wood Flame
Fuel Used -626 g
Figure 6

Figure 7 - Stoves that use less energy than the 3 Stone Fire


## Why do some stoves emit less carbon monoxide?

Propane is a clean burning fuel that produces a hot, blue flame. Propane is stored under pressure in tanks. When released, the pressure causes mixing of the gas, fire and air, resulting in very little pollution. The alcohol and kerosene stoves in this study were not pressurized and were less successful at reducing harmful carbon monoxide (CO) emissions.

As can be seen in the following graph (Figure 8), two wood-burning stoves equipped with fans were quite successful in reducing the amount of CO. Adding a small electric fan to a wood-burning stove helps in many ways. The jets of hot air create improved mixing that forces the CO to interact with air and flame, resulting in more complete combustion and dramatically reduced emissions of CO.

The Wood Gas stove shoots jets of air into and across the top of the fire, creating a zone in which fuel, air and fire are so well mixed almost complete combustion occurs. The Wood Flame stove blows air up from under the bottom of the fire. It is almost as successful as the Wood Gas stove in reducing CO. Creating a zone of mixing in or above the fire is an effective technique.

Please note that this page references total emissions (PM and CO) produced by the stove, including PM and CO emissions that would normally be exhausted from the house via a chimney not emissions to which the cook is exposed. Many chimney stoves that resulted in very low emissions in the test kitchen emitted high levels of PM and CO as measured under the collection hood from the chimney exit.

Adding an inexpensive fan to a wood-burning stove helps burn wood very cleanly. In many places where biomass fuel is used for cooking, electric power is available. In these locations, wood-burning stoves with fans seem to have a great potential to reduce both fuel use and harmful emissions. The fuel savings and health benefits should far outweigh the cost of the electricity used.

## The three T's

Carbon monoxide and particulate matter always form when fuel and air do not completely mix, and complete mixing does not occur in stoves with natural draft. The orange color of a flame comes from the radiation of particulate matter (soot) within the flame. Blue flame results from the reaction of carbon monoxide to produce carbon dioxide. So, colored flames indicate that PM and CO are reacting.

Emissions of these harmful pollutants can be reduced by burning them before the exhaust cools. Wood stove designers know that this burnout requires the three T's: time, temperature and turbulence. Time indicates that the longer the exhaust gas stays hot, the longer pollutants have to burn. Temperature indicates that the gas needs to stay as hot as possible; the reactions stop when the gas gets too cool. Turbulence is an engineering term for rough flow. If the air is turbulent, pollutants have a greater chance of coming into contact with oxygen so they can burn out.

Figure 8-Stoves that emit less CO than the 3 Stone Fire


## Which wood-burning stoves produce less particulate matter?

Many factors can decrease the emissions of unburned particles. The mixing of hot gases, air, and flame in the Wood Gas and Wood Flame fan stoves dramatically reduces particulate matter (PM) emissions. If electricity is available, biomass stoves with fans, such as propane, alcohol, and kerosene stoves, seem to have a great potential for protecting health by reducing indoor air pollution.

Charcoal-burning stoves made about one-quarter of the PM emissions compared to the 3 Stone Fire in these tests. Although charcoal can produce large amounts of CO, PM emissions were relatively low.

The Uganda 2-pot, Justa, Patsari Prototype, 20 L Can Rocket, and Onil stoves create approximately one-third to one-half the PM made by an open fire. These five stoves have low-mass rocket-style combustion chambers (Figure 9). This type of combustion chamber reduces PM and CO emissions. The VITA and Mud/ Sawdust stoves, on the other hand, are shielded-fire stoves without insulated combustion chambers and do not significantly reduce PM.

Again, adding a fan to a wood-burning stove is shown to clean up combustion. Efficient mixing is responsible for the reduction of PM in the fan and to a lesser degree in the Rocket designs.

Figure 9 - Five stoves with Rocket-style combustion chambers


Please note that this page references total emissions (PM and CO) produced by the stove, including PM and CO emissions that would normally be exhausted from the house via a chimney not emissions to which the cook is exposed. Many chimney stoves that resulted in very low emissions in the test kitchen emitted high levels of PM and CO as measured under the collection hood from the chimney exit.

Figure 10 - Stoves that emit less PM than the 3 Stone Fire


## What was the average firepower and turn-down ratio?

Firepower is a measure of how much energy is released each second. More energy is required to quickly boil water than to simmer water. The most effective cooking stove should be fuel efficient at both high and low power operation.

Figure 11 shows the average high firepower for boiling and the low firepower for simmering for each tested stove. It should be noted that in the University of California, Berkeley Water Boiling Test, the pot is uncovered, which increases the energy input needed to maintain the water at three degrees below full boil. The ratio between the high and low firepower (high firepower divided by low firepower) is called the turn-down ratio (TDR). It is a measure of how well the stove can be "turned down" from high to low power.

A TDR of 2 means that half the fuel was consumed while maintaining a simmering temperature, compared to the amount of fuel used to bring the water to boil. Cooks usually appreciate a stove that is capable of both high-and low-power operation. Many foods will burn if the firepower cannot be sufficiently decreased.

It is interesting to note that the liquid-fueled stoves were generally low powered and used nearly the same energy to boil and simmer food. The Mud/ Sawdust (TDR 3.9) and VITA (TDR 3.8) stoves had the highest TDR. The average for the other wood-burning stoves without chimneys was 2.4. The average for stoves with chimneys was 2.2 . The Gyapa charcoal stove (TDR 2.8) scored slightly higher. While TDR seems to be an important stove characteristic, the graphs on the following page (Figures 12 and 13) indicate that TDR alone does not predict fuel efficiency.

Figure 11- Firepower and turn-down ratio of stoves


Figure 12- Energy to cook vs. turn-down ratio


Figure 13- Firepower of stoves and energy to cook 5 L


## What is the effect of adding a chimney to a wood-burning cook stove?

Chimneys protect the cook and family from smoke. The chimney has evolved over time to be the primary solution to indoor air pollution. If the stove and chimney do not leak, pollution is removed from inside the house. In these experiments, chimneys protected the testers from the dangerous levels of indoor air pollution made by fire. A functional chimney can remove essentially all the emissions made inside a stove, if the smoke does not leak into the room.

Figures 16 and 17 compare the performance of stoves with and without chimneys. Chimneys removed all but $1 \%$ of the CO and PM from the test kitchen. The pollutants that did enter the kitchen escaped through small leaks in the stove.

The stoves with chimneys in this study were slower to boil water and used more wood to boil and then simmer water (Figure 14). However, these stoves were mostly griddle stoves in which hot gases transfer heat through a heavy metal surface to the pots or food placed directly on the griddle. It was the griddle that caused these differences, not the chimney.

Figure 14-Comparison of nonchimney and chimney stoves

|  | Average <br> No Chimney | Average <br> Chimney |
| :--- | :--- | :--- |
| Time to Boil | 19 min | 33 min |
| Fuel to Cook | 870 g | $1,400 \mathrm{~g}$ |
| CO in Kitchen | 340 ppm | 3 ppm |
| PM in Kitchen | $18,000 \mu / \mathrm{m}^{3}$ | $280 \mu / \mathrm{m}^{3}$ |

Griddle stoves such as the Justa, Onil and Ecostove have a great advantage in that food can be cooked directly on the hot surface. The griddle stove is necessary and popular in places where flat breads are cooked. In Central America tortillas are a staple food. As the tortillas are made, a pot of beans often simmers to completion at the back of
the griddle. The griddle stoves in this study all had chimneys that removed essentially all emissions out of the kitchen.

As can be seen in Figure 17, stoves without chimneys often created dangerous levels of pollution in the test kitchen.

The Uganda 2-pot stove (Figure 15) is the only stove equipped with a chimney studied in which pots are submerged into the stove body. It does not have a griddle. Instead, the hot gases flow against the bottom and sides of the two submerged pots, which fit tightly in holes that prevent smoke from escaping into the kitchen. As can be seen in Figure 16, the Uganda 2-pot chimney stove boils water as quickly and uses about the same amount of fuel as stoves without chimneys. Stoves without chimneys are shown on the left side of the graph, while those with chimneys are on the right.

Lacking a sealed griddle, the Uganda 2-pot stove leaks more pollution into the room than do other stoves equipped with chimneys. However, the levels of indoorair pollution are greatly reduced compared to a 3 Stone Fire. If the stove had better seals around the pot, more of the smoke would exit the chimney.


Figure 15

Figure 16 - Chimney stoves:fuel to cook and time to boil


Figure 17 - Chimney stoves: kitchen emissions to cook 5 L


## How does ventilation affect pollution in a kitchen?

To make all the tests as similar as possible, the test kitchen doors and windows were closed when the stoves were being tested and testing was done only on calm days. If the wind was blowing one day and not the next, the levels of CO and PM measured in the building would be affected, making accurate comparisons difficult. Tests were conducted to determine whether opening the door or window, or making a small hole in the roof, would significantly reduce the indoor air pollution. The tests described here explore this question.

In this study, 20 Kingsford charcoal briquettes were burned in the approximately 15 -cubic-meter test kitchen with approximately 3 air exchanges per hour. The emissions-monitoring equipment consisted of six HOBO carbon monoxide monitors and two Airmetrics Minivols pump and filter particulate meters. The Minivol draws 5 L of room air per minute though a filter that collects PM2.5 (particles less than 2.5 micrometers in aerodynamic diameter).


[^0]Figure 18 - Test kitchen diagram

Three tests were performed for each configuration:

1. The window, door and hole in the roof closed.
2. The 0.6 by 1.8 m door open.
3. A 20 by 25 cm hole in the roof open.
4. A 28 by 36 cm window, along with the 20 by 25 cm hole in the roof, open.

The kitchen diagram (Figure 18) shows the location of openings as well as the placement of monitoring equipment.

The charcoal was left to burn vigorously for 30 minutes. It was then quickly removed through a small opening, which was then closed. The test continued for another 30 minutes as levels of CO and PM declined.

Figure 19 shows the peak concentration of CO reached after the half hour of burning, the average CO level throughout the test, and the average concentration of PM during the four levels of ventilation. As Figure 19 shows, increasing amounts of ventilation significantly lowered levels of both types of emissions.

Figure 20 summarizes the variability and potential reduction in indoor air pollution resulting from the four configurations. The levels of both CO and PM with the door and window closed were elevated, as can be expected. Opening the door was highly effective in this study, reducing emissions by $96 \%$.

Making a small hole in the roof also significantly improved air quality. However, simultaneously

## Kitchen Dimensions:

$10 \mathrm{ft}(3 \mathrm{~m})$ wide X $8 \mathrm{ft}(2.4 \mathrm{~m})$ deep X $6 \mathrm{ft}(1.8 \mathrm{~m})$ high X $8 \mathrm{ft}(2.4 \mathrm{~m})$ peak
Door: $2 \mathrm{ft}(0.6 \mathrm{~m}) \mathrm{X} 6 \mathrm{ft}(1.8 \mathrm{~m})$
Window: 11 in ( 0.28 m ) X 14 in ( 0.36 m )
Hole in Roof: 9.8 in ( 0.25 m ) X 7.9 in ( 0.2 m )
Stove height: 2 ft . $(0.6 \mathrm{~m}$ )

Figure 19-CO and PM in the test kitchen with differing ventilation

opening a small window did little to further reduce levels of pollution, possibly because the window did not add much flow to the movement of CO and PM through the smoke hole in the roof.

Increasing ventilation seems to be an effective strategy for decreasing indoor air pollution in
houses in which biomass fuel is burned. Increasing ventilation dramatically reduced both CO and PM in the test kitchen. Opening the door was especially effective. Cutting a small, covered hole in the roof also removed most of the smoke from the kitchen because the smoke collects near the ceiling in a room.

Figure 20-CO and average PM level reduction by ventilation

|  |  | Average | \% Reduction <br> from Closed <br> Kitchen | Expected IAP <br> Reduction for <br> This Ventilation |
| :--- | :--- | :--- | :--- | :--- |
| Closed Kitchen | CO Average (ppm) | 54 |  |  |
|  | CO Peak (ppm) | 160 |  |  |
|  | PM Average $\left(\mathrm{ug} / \mathrm{m}^{3}\right)$ | 1,025 |  |  |
| Hole in Roof | CO Average $(\mathrm{ppm})$ | 18 | $67 \%$ |  |
|  | CO Peak $(\mathrm{ppm})$ | 41 | $75 \%$ |  |
|  | PM Average $\left(\mathrm{ug} / \mathrm{m}^{3}\right)$ | 334 | $67 \%$ | $70 \%$ |
| Window and Hole in Roof | CO Average $(\mathrm{ppm})$ | 14 | $75 \%$ |  |
|  | CO Peak $(\mathrm{ppm})$ | 44 | $73 \%$ |  |
|  | PM Average $\left(\mathrm{ug} / \mathrm{m}^{3}\right)$ | 345 | $66 \%$ |  |
| Door Open | CO Average $(\mathrm{ppm})$ | 1 | $97 \%$ |  |
|  | CO Peak $(\mathrm{ppm})$ | 6 | $96 \%$ | $96 \%$ |
|  | PM Average $\left(\mathrm{ug} / \mathrm{m}^{3}\right)$ | 66 | $94 \%$ |  |

Natural ventilation is driven by air pressure due to differences in air density. If indoor air is warmer than outdoor air, the flow out of the hole in the roof can be increased. To some extent, this stack effect depends on winter and summer temperatures.

## Stratification of CO and PM in the test kitchen (See Figure 21)

Three additional tests were run to study stratification in the closed kitchen using six HOBO CO data loggers and six Minivol PM monitors at three different heights on opposite sides of the room. The HOBOS and Minivols were located across from each other at 1 meter, 1.4 meters and 1.8 meters above the floor.

Both CO and PM stratified by height in the test kitchen, collecting densely at the ceiling and decreasing gradually towards the floor. Levels were lowest nearest the floor, suggesting that exposure could be reduced by sitting instead of standing while cooking. Some horizontal stratification was also observed.

Figure 21-CO and PM concentrations by height across the unventilated kitchen


## How do fans improve woodburning cook stoves?

As can be seen in Figures 22 and 23, wood-burning cook stoves equipped with fans have several advantages. In natural-draft stoves, smoke, air and flame are not forced to mix; smoke can go in one direction and flame can go in another. The smoke can easily escape combustion, so CO and PM emissions are often high. The averages of CO and PM in the test kitchen and under the emissions hood are dramatically reduced when a fan is used.

In the Wood Gas fan stove (Figure 24), jets of air are blown into and over the fire. The Wood Flame fan stove (Figure 25) blows jets of air only into the fire from under the floor of the combustion chamber. The two stoves used an average of 540 grams of wood to boil and then simmer 5 L of water for 45 minutes. The average stove without a fan used 870 grams of wood to accomplish the same task. The velocity of hot gases and possibly gas temperature are increased by the jets of air. Radiation to the pot can also be increased in a fan stove because the distance between the fire and the pot is usually reduced. For these reasons, the heat transfer to the pot is increased and less wood is needed for cooking.

Although the hot gases contact only the bottom of the pot in the Wood Flame and Wood Gas stoves, the fuel used to boil and simmer water is less than the VITA, Mud/Sawdust and 20 L Can Rocket stoves, even though these natural-draft stoves all force the hot gases to flow against the sides of the pot after contacting the bottom. The fan increases both heat transfer and combustion efficiency.

Stoves with fans are remarkably clean burning (see Figure 26). Even though the 20 L Can Rocket stove is considered a "clean-burning" wood stove, the fan stoves are much cleaner because the production of PM and CO is considerably reduced. A stove equipped with a chimney or a fan can reduce emissions and exposure to pollutants while cooking with wood.

Figure 22 Fan stoves fuel to cook and time to boil 5 L


Figure 23 Fan stoves CO and PM emissions to boil 5 L


Figure 24


Figure 25


Figure 26 Comparison of stoves with and without a fan

|  | Average of <br> No Fan Stove | Average of <br> Fan Stove | Fan/No Fan |
| :--- | :--- | :--- | :--- |
| Time to Boil (min) | 19 | 22 | $117 \%$ |
| Fuel to Cook (g) | 870 | 540 | $63 \%$ |
| CO in Kitchen (ppm) | 340 | 90 | $27 \%$ |
| CO under hood (ppm) | 43 | 8 | $19 \%$ |
| PM in Kitchen (ug/m $)$ | 18,000 | 2,200 | $12 \%$ |
| PM under hood $\left(\mathrm{ug} / \mathrm{m}^{3}\right)$ | 2,500 | 37 | $1 \%$ |

## How do wood- and charcoalburning stoves compare?

The charcoal used in this study was made in Mexico from the trunks and branches of mesquite trees. Two charcoalburning stoves were tested, one from Mali (Figure 27) and one from Ghana (Figure 28). In these tests fuel use and emission measurements began 10 minutes after the charcoal was lit. The two stoves were found to be more effective than traditional charcoalburning models studied in previous tests.

Charcoal is made by heating wood or other biomass fuel inside a relatively air-tight enclosure, such as an earth covered pit in the ground. Smoke escapes through holes in the covering
and causes air pollution. In this case, wasted smoke is fuel that could have been used to cook food. However, there are more efficient methods of producing charcoal that can avoid energy losses. Examples include producing charcoal in stoves that burn the volatiles in biomass to produce heat for cooking and producing charcoal from crop residues that otherwise would be burned. Between 70\% and $80 \%$ of the energy in wood is used to produce charcoal. 5 "The charcoal thus produced retains the same shape of the original wood but is typically just one-fifth the weight, one-half the volume, and one-third the original energy content." ${ }^{\text {6 }}$

Figure 29 compares the energy in charcoal and wood fires. Since so much energy is lost when making charcoal, wood stoves were much more fuel efficient. Almost three times as much total energy was used to cook food with the charcoal stoves in these tests.

As can be seen in Figure 30, the two charcoal stoves boiled water slower than the 3 Stone Fire and the average of all single-pot, wood-burning stoves. Charcoal seems especially well suited to simmering, but is somewhat low powered for rapid boiling.

[^1]Figure 29- Charcoal comparison: Energy used to cook 5 L


The great advantage of charcoal is that it continues burning at a steady rate, without the need to constantly feed the fire, as in a wood-burning stove. Reducing the air entering the fire prolongs the useful cooking time and provides a gentle heat suited to simmering.

Charcoal is known to produce a large amount of CO. In these tests was certainly true. Charcoal stoves produced at least twice as much CO as any other stove. On the other hand, PM from charcoal-stove emissions was low, especially during simmering. The significant reduction in PM when using charcoal could help reduce human health impacts, except that CO emissions are so high (Figure 31).

Figure 30 - Charcoal comparison:Time to boil 5 L


Figure 31 - Charcoal comparison: CO and PM emissions to cook 5 L


## How does a retained heat cooker help when cooking?

When food simmers, the fire replaces the constantly lost heat from the pot. If the heat were not lost but captured instead, then less fuel would be needed for cooking. Placing the pot of boiling food in an insulated container keeps the food hot enough to simmer it to completion. In the same way, a drafty and uninsulated house has to have a big fire in the heating stove going all the time to keep the house warm. Even if no fire is lit, the super-insulated, almost airtight house can stay warm for a long time.

After a pot of food boils, the contents are close to $100^{\circ} \mathrm{C}$. When the hot pot is placed in a superinsulated, almost airtight box, the food finishes cooking, because the stored heat stays in the food. Once the pot is in the box, food cooks without further attention. The retained heat cooker (RHC) or Haybox as it is called in some parts of the world, saves time, effort, and fuel, freeing the cook from long hours of watching the slow fire when simmering food.

Figures 32 and 33 depict both time and fuel savings when using a retained heat cooker to simmer food. Approximately $50 \%$ savings in both categories can be expected.

Figure 32 - Use of the RHC potentially saves time tending the stove during simmering


Figure 33 - Use of the RHC potentially saves fuel during simmering


Because the fuel is initially used only for boiling food, cooking with an RHC creates much less pollution, helping to clean up the air in the kitchen. In these tests, using an RHC reduced, on average for all stoves, CO emissions by $56 \%$ and PM emissions by $37 \%$ (Figures 34 and 35).

RHCs have been used for hundreds of years. They can save time and effort which can be devoted to
other tasks. The attraction of the RHC begins with its convenience. The fuel savings and decrease in harmful emissions add to the benefits of retained-heat cooking. More information on Retained Heat Cookers can be found in PCIA's Guide to Designing Retained Heat Cookers available at www.PCIAonline.org/resources.

## Important Note!

Food should be boiled for at least 5 minutes to kill bacteria before being placed in a retained-heat cooker.

Figure 34 - Potential PM emission savings using an RHC during simmering


Figure 35 - Potential CO emissions savings using an RHC during simmering


## What is efficiency?

People are naturally drawn to the word "efficiency" and think that improved thermal efficiency means decreased fuel use when cooking food. Unfortunately, choosing a stove based on thermal efficiency can result in the selection of a stove that is not necessarily as fuel saving as possible.

Thermal efficiency is a measure of how much energy in the wood fuel is transferred into the cooking pot. Because there is no good way to measure this heat transfer, it is often approximated by measuring the amount of water evaporated; but this technique does not indicate how much of that energy is useful for cooking. Boiling off a lot of extra steam can result in a higher "efficiency" number, but it will not cook food any faster than a more moderate rate of simmering.

A water-boiling test is usually used to determine efficiency. There are many versions of water-boiling tests. Varying test methods result in numbers for efficiency which are not readily comparable. An alternative approach called "specific consumption" replaced efficiency in the 1985 VITA International Testing Standard.

Specific consumption is the fuel used per unit of product produced. The unit of product could be bowls of cooked food and or loaves of bread. In this case, liters of boiled and simmered water represent cooked food. Remember that we are talking about the weight of finished product (cooked food, or in this case, water remaining at the end of the test), not starting weight (uncooked food, or in this case, water at the beginning of the test).

Figures 36 and 37 rank the energy and fuel used by different stoves to do the same task (producing a liter of boiling water, then simmering it for 45 minutes). The efficiency of the stoves is represented by the line. It can be seen that the two measures of stove performance are not closely related. "Thermal efficiency" rewards the production of excess steam, while "specific consumption" penalizes it. Making excess steam results in less final product and is not needed for fuel-efficient cooking. The VITA 1985 International Testing Standard recommends "Specific Consumption" as the more reliable indicator of stove performance.

Figure 36 - Comparison of specific energy consumption and thermal efficiency to boil 1 L


Stove power must be sufficient to overcome heat losses through the sides of the pot and to supply the heat required for vaporization of water. As the water approaches the boiling point, more power is needed to offset heat losses from the pot. This high-energy requirement is difficult for low-powered stoves to meet, and they remain in the pre-boiling state longer than high-powered stoves. At near-boiling temperatures, a lot of water evaporates. For this reason, low-powered stoves can evaporate more water than high-powered stoves before they reach a boil. However, this condition may not be efficient because the stove is struggling to reach the boiling point.

The requirement for more energy as the boiling point is approached creates an energy "hump" which low-powered stoves take longer to overcome. The low-powered stove boils off a great deal of water because the water remains in the high steammaking condition longer than the higher powered stove. This condition results in long times to boil and large losses of water through vaporization. Increased steam production can produce high efficiency numbers even though fuel is being used for a longer period.

Problems with efficiency become even more evident when simmering water. Simmering attempts to maintain hot water (or food) at just under the boiling temperature, using the minimum amount of fuel. The most effective methods for simmering water (such as the use of pot lids, insulation, retained-heat cookers, etc.) cannot be measured by the method of estimating heat transfer from steam loss.

Problems with thermal efficiency have been recognized for decades. Thermal efficiency in conjunction with power output (at high and low power) can be used to make accurate predictions about stove performance. By using the two factors together and defining a cooking process (cooking rice, for example), one can calculate cooking time, fuel use, water loss and so forth. However, thermal efficiency by itself is not a reliable predictor of performance and should only be used with other measures, such as specific consumption, when comparing cook stoves.

Figure 37 - Comparison of specific energy consumption and thermal efficiency to simmer 1 L


# Does increasing heat transfer efficiency have to decrease combustion efficiency? 

Dr. Grant Ballard-Tremeer ${ }^{7}$ and Dr. Kirk Smith have pointed out that getting more of the heat from a fire into the pot can also result in more pollution. For example, lowering a pot closer to the fire results in lower fuel use but also makes more smoke. Smith summarized this observation as follows:
"Combustion efficiency (CE) may not be worth pursuing from an overall efficiency (OE) standpoint, but is very much worth pursuing from a pollution standpoint because pollution emissions are a direct function of (1-CE). Thus, a relatively slight lowering of CE, which may produce only a slight change in OE, can produce substantial increases in pollution, even on a per meal basis." ${ }^{8}$

Smith lists examples from his studies where small decreases in combustion efficiency, following changes to increase heat transfer efficiency, resulted in two to three times more pollution per meal. Is it always true that getting more of the heat from a fire into the pot results in poorer combustion and more smoke?

Dr. Larry Winiarski ${ }^{9}$ approached designing stoves by separating functions along the same lines as Ballard-Tremeer and Smith. His hope was that if wood were burnt in an improved combustion chamber, cleaner hot gases could be forced to flow against the pot without making more smoke. Winiarski hoped that if CE was close to $100 \%$, improving heat transfer efficiency (HTE) would not decrease combustion efficiency.

Figure 38 shows examples where increasing HTE does decrease CE. Emission factors are useful to compare because emission factors report the mass of pollution per mass of wood burned, indicating the cleanliness of combustion. As can be seen, the Mud/Sawdust stove and especially the VITA stove sacrifice clean burning for reduced fuel use and quicker time to boil. However, neither of these stoves has an improved combustion chamber.

In the VITA and Mud/Sawdust stoves, the fire is surrounded by a metal or earthen wall and moved closer to the pot. In both stoves, small channels force the hot gasses to also flow against the sides of the pot. This type of stove can make more pollution per meal because it does not address combustion efficiency.

Figure 38 - Comparison of specific energy consumption and thermal efficiency to simmer 1 L

|  | 3 Stone Fire | Mud/Sawdust | VITA | 20 L Can Rocket |
| :--- | :--- | :--- | :--- | :--- |
| Time to Boil $(\mathrm{min})$ | 27 | 16 | 14 | 22 |
| Fuel to Cook $(\mathrm{g})$ | 1,100 | 780 | 690 | 730 |
| CO to Cook $(\mathrm{g})$ | 56 | 49 | 43 | 15 |
| PM to Cook $(\mathrm{g})$ | 2,400 | 2,400 | 2,200 | 1,300 |
| Emission Factor CO $(\mathrm{g} / \mathrm{kg})$ | 51 | 43 | 93 | 14 |
| Emission Factor PM $(\mathrm{mg} / \mathrm{kg})$ | 3,500 | 5,100 | 8,300 | 2,000 |

[^2]Winiarski's 20 L Can Rocket stove, on the other hand, has an insulated combustion chamber that cleans up smoke before it can escape. This feature can simultaneously improve combustion efficiency and heat-transfer efficiency. Fuel use and emissions are both reduced.

As can be seen in Figure 39, emission factors in the 20 L Can Rocket stove are reduced, compared to a carefully made 3 Stone Fire. Well-engineered combustion chambers in cooking stoves create cleaner gases which can be forced to more effectively get heat into the pot. This type of stove can use less wood and make less smoke, while boiling water faster than the 3 Stone Fire.

Figure 39 - Comparison of emission factors (EF)


## Does CO predict PM?

It might possible to use measured CO levels to predict a general expected level of PM for a given fuel. Generally, it is much easier to measure CO than PM. If a correlation can be established, stove researchers might be able to simplify measurements in the field.

Some researchers report that CO and PM are related. The levels in a house or in the streets of a city may generally follow a similar pattern. However, CO and PM do not rise and fall together as combustion occurs.

CO is created by gases that are not burnt up in the flame. Burning wood usually produces high levels of CO when the fire is started. CO levels rise as burning wood makes charcoal. Alternately, PM is seen when the fire makes flame. Both CO and PM tend to rise when fresh wood is added. Each pollutant is produced by a different mechanism, at various times during a cooking task. Figure 40 shows a typical record of CO and PM emissions during a Water Boiling Test.

In the tests done at Aprovecho, charcoal- and liquid-fuel stoves did not emit levels of CO and PM like wood-burning stoves. Burning charcoal makes high levels of CO, but relatively low levels of PM. Liquid gas fuels produce almost no PM when the stove is properly tuned.

As can be seen in Figure 41, in the test kitchen the levels of CO and PM when water is brought to a boil and simmered for 30 minutes do seem to be related. Most stoves that emitted less CO also produced less PM.

Figure 42 details the same comparison for stoves tested under the emissions collection hood. It should be noted that due to problems with the data, only one Water Boiling Test for PM could be used. The PM hood results show the average of cold and hot starts to boil that are added to the emissions for simmering for 45 minutes. The levels of CO are the average of three Water Boiling Tests. This analysis also shows a positive relation, although the Ghana Wood and Ecostove PM levels seem unexpectedly high.

Figure 40 - Comparison of emission factors (EF) for improved heat transfer


In this study, the levels of CO and PM do seem to be related for stoves burning the same type of fuel. Clean-burning stoves remove most of the PM and CO, and polluting stoves emit high levels of PM and CO. Stoves with chimneys remove both PM and CO from the kitchen. However, some stoves reduce CO while increasing PM. Of course, charcoal stoves emit much more CO than PM.

It may not be safe to assume that clean-burning stoves reduce both PM and CO in proportional amounts, because this assumption does not hold true for all stoves. Further studies would be required for a particular application.

Figure 41 - Relation of CO and PM for a cooking task in the test kitchen


Figure 42 - Relation of CO and PM for a cooking task under the emissions collection hood


## How do hydrocarbon emissions compare?

Hydrocarbons are unburned gases with hydrogencarbon bonds such as propane, methane, butane and hexane. Like other pollutants, hydrocarbons are harmful to human health and contribute to global warming. The Enerac 3000E (Figure 43), used in the Aprovecho emissions hood, uses nondispersive infrared (NDIR) to count the number of carbon bonds to determine the concentration of hydrocarbons, reporting them as propane. Since the Enerac is designed for higher concentrations of hydrocarbons and counts all as propane, the results presented in Figure 44 may not be accurate in magnitude. However, it is possible to compare the relative amounts of hydrocarbons emitted by stoves.

As measured by the Enerac 3000E, the two charcoal stoves emitted about twice as much unburnt hydrocarbons as the wood-burning stoves. As with CO and PM emissions, the wood-
burning stoves that produced the least amount of hydrocarbons used either rocket combustion chambers or fans. Because there is significant difference in measured hydrocarbon emissions, further study seems warranted.


Figure 43 - Enerac 3000E

Figure 44 - Hydrocarbon emissions to cook 1 L


## How does emission testing with a hood or in a test kitchen compare?

Two different methods were used in this study to measure the emissions from different stoves when performing a standard task: boiling and simmering water in the same pot.

The exhaust collection hood (Figure 45) creates a constant flow of air which is carefully measured so that the amount of pollutants is known. The amount of air leaving the test kitchen (Figure 46) also has to be controlled. Tests cannot be done on windy days and all windows and doors must be closed as the stove burns fuel. The intent of both methods is to reduce the factors that affect stove performance measurements.


Figure 45 - Testing hood
The hood collects all the smoke and draws it past measuring devices; monitoring equipment in the test kitchen is immersed in the smoky air. One of the big differences between testing with a hood or in a kitchen is the cost. The equipment used in the hood (Enerac 3000E, Radiance M903 Nephelometer, etc.) cost more than $\$ 20,000$. The instruments for the hood were chosen to provide real-time information about emissions so that designers could understand stoves better. It also provides specific emissions in pollutant per cooking task or per kilogram of fuel burned. The portable equipment used in the test kitchen (AP Buck filter system, HOBO CO
monitor) cost less than $\$ 2,000$. However, it cannot provide such detailed information.

It took more than a year to build the hood and calibrate it so that the results were usable. The portable equipment used in the test kitchen is made to be used by field personnel with little training. One reason for the long development time for the hood was that few stove developers had used this kind of system before. New procedures and instruments had to be developed. On the other hand, many researchers had already been using infield indoor air pollution (IAP) monitors which were placed in the test kitchen to assess air quality.

How did the data from the hood and the test kitchen compare?
Three Water Boiling Tests (WBTs)were performed under the hood and three shortened WBTs were performed in the test kitchen. The average results for each stove are shown in Figures 47 and 48. The stoves equipped with chimneys removed almost all pollutants from the test kitchen. The emissions of the stoves with chimneys were measured under the hood from the chimney exit. For this reason, the results from the hood and in the test kitchen

It should be noted that testing stoves in a test kitchen exposes the tester to high levels of smoke and carbon monoxide. In these tests, the stove operator always wore a respirator that directly provided fresh air for breathing.


Figure 46 - Test kitchen
for stoves with chimneys cannot be compared. The kitchen test shows the levels of pollutants in the room. The hood tests show total emissions that affect the environment.

In most cases, for stoves without chimneys, the levels of CO measured under the hood and in the test kitchen were quite similar.

Figure 48 compares the PM data from the hood and test kitchen. Unfortunately, the PM data from two of the three tests done under the hood could not be used because of technical problems. However, there is general agreement (for stoves without chimneys) between the results for PM from the test kitchen and the hood. Again, the
chimney stoves on the right side of the graphs were measured differently and cannot be compared.

When stove prototypes are being developed, emission data are important. Measuring how cleanly stoves operate is necessary for the evaluation of stoves that are to be distributed.

Using an emission hood or a standardized test kitchen are two ways to provide data on pollution made by stoves. The test kitchen has the advantage of lower cost and easier to use equipment. The hood is more accurate and provides more reliable information that answers a wider variety of questions. Either, when used carefully and systematically, can be used to compare cook stoves.

Figure 47 - CO to cook 5 L under emissions hood and average CO Level in test kitchen


Figure 48 - PM to cook 5 L under emissions hood and average PM Level in test kitchen


## What is an "improved" cook stove?

The eighteen stoves in this study were tested under an emission hood and in a test kitchen, using various monitoring devices. Capturing the emissions in the hood makes it possible to estimate the mass of CO and PM made during a cooking task. In the test kitchen, the parts per million (ppm) of CO and the micrograms per cubic meter $\left(\mu \mathrm{g} / \mathrm{m}^{3}\right)$ of PM in the room air are monitored, using portable equipment. Assuming that the air exchanges in the test kitchen are relatively constant, higher readings of pollution in the air are caused by stoves that are burning less cleanly.

The $\mathrm{CO} / \mathrm{CO}_{2}$ ratio has been suggested as another method for determining how cleanly a stove is
burning. It is calculated by dividing the amount of CO by the amount of $\mathrm{CO}_{2}$. A lower ratio means that more $\mathrm{CO}_{2}$ and less CO were produced during the Water Boiling Test. If biomass fuel is burned cleanly, more $\mathrm{CO}_{2}$ is made and less CO is emitted. The CO is combusted and changed into $\mathrm{CO}_{2}$. A stove that is operating at $100 \%$ combustion efficiency would emit only $\mathrm{CO}_{2}$ and water.

It may be possible to use the $\mathrm{CO} / \mathrm{CO}_{2}$ ratio as a benchmark for stove combustion efficiency. The South African Bureau of Standards suggests that the $\mathrm{CO} / \mathrm{CO}_{2}$ ratio from paraffin (kerosene) stoves should be $2 \%$ or less. Both CO and $\mathrm{CO}_{2}$ are relatively simple to measure with equipment that has a combined cost of about $\$ 600$.

The following graph (Figure 49) shows the average results of three Water Boiling Tests of each stove

Figure 49 - Comparison of $\mathrm{CO} / \mathrm{CO}_{2}$ ratio and CO produced per liter to cook

conducted under the emissions hood. The propane, kerosene, fan stoves and the rocket stoves meet the suggested benchmark of $2 \%$ of $\mathrm{CO} / \mathrm{CO}_{2}$. The $\mathrm{CO} /$ $\mathrm{CO}_{2}$ ratio also seems to be correlated to another measure of CO generated from the Aprovecho emissions hood: the amount of CO produced to boil and simmer 1 L of water. The emissions hood results are presented as the mass of CO produced per liter of water boiled and then simmered during the Water Boiling Test.

Benchmarks for emissions can also be created using the data from the hood or test kitchen. Fuel use is also comparable. Figure 50 shows two lines drawn across the graph that could establish a proposed level of acceptable performance.

The Shell Foundation asked Aprovecho Research Center to use the data from these tests to create proposed benchmarks to encourage the production of improved cooking stoves (ICS) that save fuel and reduce indoor air pollution. The lines that cross Figures 50, 51 and 52 are for stoves with and without chimneys. Both fuel used and energy used to cook 5 L are included.

A suggested benchmark for fuel and energy use is (Figure 50):

1. Fuel use: Using the International Testing Pot, a cooking stove without a chimney should use less than 850 grams of wood or less than $15,000 \mathrm{~kJ}$ of energy to bring to boil 5 L of $25^{\circ} \mathrm{C}$ water and then simmer it for 45 minutes during the University of California, Berkeley revised Water Boiling Test. Stoves equipped with chimneys should accomplish the same task, consuming less than 1,500 grams of wood or $25,000 \mathrm{~kJ}$ of energy.

## A suggested benchmark for CO produced is

 (Figure 51):2. Emissions: A cooking stove without a chimney should produce less than 20 grams of carbon monoxide to boil 5 L of $25^{\circ} \mathrm{C}$ water and then simmer it for 45 minutes during the University of California, Berkeley revised Water Boiling Test. Wood-burning stoves equipped with chimneys are exempt from the above standard if the stove does not allow more than 50 ppm of CO to pollute the air within 30 cm of the stove in the standard test kitchen with a controlled air exchange.

Figure 50 - Fuel and energy to cook 5 L vs. benchmark


## A suggested benchmark for PM is (Figure 52):

3. Emissions: A wood-burning stove without a chimney should produce less than 1,500 milligrams of PM (with a total size of 2.5 micrometers in aerodynamic diameter or smaller) to boil 5 L of $25^{\circ}$ C water and then simmer it for 45 minutes during the University of California, Berkeley revised Water Boiling Test.

These preliminary suggestions show how performance benchmarks can be created from data generated from various methods. Benchmarks can
be developed using the $\mathrm{CO} / \mathrm{CO}_{2}$ ratio or from test kitchen or emission hood results. Each method has advantages and disadvantages. $\mathrm{CO} / \mathrm{CO}_{2}$ and test kitchen results are obtained using less expensive but less accurate equipment. The emission hood data are probably the most accurate, but a hood system is complicated and more expensive. Before a set of performance benchmarks is generally adopted, more research and development are needed. The suggested fuel use and emission levels can be adjusted up or down. Benchmark levels can be determined using various emission monitoring systems.

Figure 51 - CO to cook 5 L and average CO level in test kitchen vs. benchmark


Figure 52 - PM to cook 5 L and average PM level in test kitchen vs. benchmark


## How can wood-burning cook stoves be improved?

The success of some of the groups of stoves in this study point out a few simple techniques that help to improve performance.

(!)
Functional chimneys can address the problem of indoor air pollution. Chimneys are the practical solution that evolved in all developed or industrialized countries to remove harmful pollution from the indoor environment. The Onil stove, the Ecostove, and the Uganda 2-pot stove (Figure 53) have chimneys that removed most emissions from the test kitchen.

The test kitchen is a $15 \mathrm{~m}^{3}$ building in which a door and a window are closed to simulate the worst conditions when fire is used inside in a cold climate. Even in this mostly unventilated structure, stoves with chimneys removed most of the pollution. It is important to use a cooking stove with good draft, however. If smoke can flow out of the fuel entrance, or leak in other ways into the room, harmful emission levels will rise.

Chimney stoves dramatically reduced the emissions of PM and CO, as can be seen in Figure 54. The use of chimneys is probably the most cost-effective technique to address the problem of indoor air pollution.

Figure 53 - Chimney Stoves


Figure 54 - Concentration of CO and PM in the test kitchen generated by stoves with chimneys


(!)
Providing an insulated combustion chamber around and above the fire creates better mixing of gases, flame and air, which helps to boil water faster, reduces fuel use, and decreases CO and PM. The 20 L Can Rocket, the Uganda 2-pot, Justa, and the Patsari Prototype stoves have "rocket type" insulated combustion chambers (meaning L-shaped insulated combustion chambers)(Figure 55). The higher temperatures and improved mixing in an insulated enclosed space above the fire reduces harmful emissions (Figure 56).


Figure 55 - Five stoves with rockettype combustion chambers


Figure 56- Insulated combustion chamber CO and PM emissions to cook 5 L


(!)
Forcing the hot gases to flow against as much of the pot or griddle as possible improves heat transfer. This is an effective method to reduce the fuel needed for cooking. The 20 L Can Stove, the VITA stove, the Uganda 2-pot stove and the Mud/ Sawdust stove use small channels that direct the hot gases to contact the sides and bottom of the cooking pot. Baldwin and Winiarski have shown that improving heat transfer significantly decreases

Cylinders Surrounding the Pot - Fuel to Cook 5 L


Figure 57 fuel use.

The VITA (Figure 59) and Mud/Sawdust (Figure 58) stoves are cylinders surrounding the pot, creating a small gap between the pot and stove body. This simple technique dramatically reduces fuel use (Figure 57). In outdoor cooking situations where fuel efficiency, not reduction of emissions, is most important, this approach provides a low-cost method for decreased fuel consumption.


Figure 58


Figure 59

(7)
Stoves can be designed with small fans that create high-velocity, low-volume jets of air that mix fuel, air and flame. This mixing is mostly missing in stoves without fans. Mixing dramatically reduces pollution (Figure 61). The Wood Flame and Wood Gas stoves burn wood much more cleanly. Adding low-cost fans to stoves could provide another low-cost solution to cleaner, more efficient cooking with biomass (Figure $60 \& 61$ ).


Figure 60


Figure 61

## Appendix A

## Glossary

Benchmarks: Suggested measures of performance that, in this case, seek to define an improved cook stove.

Boundary layer: The very thin layer of slowly moving air immediately adjacent to a pot surface that insulates the pot from the hot flue gases and decreases the amount of heat that enters the pot.

Carbon monoxide: An odorless, colorless gas that is harmful to health produced by the incomplete combustion of fuel.

Convection: The heat transfer in a gas or liquid by movement of the air or water.

Combustion chamber: The area of a stove where the fuel is burned.

Combustion efficiency: The percentage of energy in fuel that is turned into heat.

Constant Cross Sectional Area: Maintaining spaces with the same volume measured at right angles to the flow throughout a stove.

Draft: The movement of air through a stove and up the chimney.

Emissions: Byproducts from the combustion of fuel that are discharged into the air.

Emissions Hood: An instrument that captures and measures the mass of emissions from burning fuels.

Excess Air: Air used for combustion that exceeds the theoretical (stochiometric) amount needed.

Firepower: A measure of how much energy is released from burning fuel per unit of time.

Flue Gas: The hot gas from burning fuel that flows up from the combustion chamber.

Grate: A framework used to hold the fuel above the combustion chamber floor.

Heat Transfer Efficiency: The percentage of available energy released from the fuel that entered the pot.

High Mass Stove: A stove made from dense materials such as earth, clay and sand that absorb heat from a fire more readily than lighter, more insulative materials.

Hydrocarbons: A mixture of gases including propane, methane and butane released from wood fuel but that remain unburnt and exit the stove due to incomplete combustion.

Mixing: The combining of air, hot gases and flame to reduce emissions. Biomass stoves do not mix air, hot gases and flame very well, so smoke and unburnt gases are often not fully combusted.
$\boldsymbol{\mu g} / \mathbf{m}^{3}$ : Micrograms per cubic meter, the measure of concentration of particulate matter in air.

Overall Efficiency: The combination of heat transfer efficiency and combustion efficiency expressed as a percentage.

Particulate Matter: The fine particles that make up smoke. They can vary in size and composition and are harmful to health when breathed. The smaller the particle, the more deeply into the body it can travel.

Pot Skirt: A cylinder, usually made from sheet metal, that creates a narrow channel around the sides of a pot to increase heat transfer efficiency.
ppm: Parts per million, a measure of the concentration of a gas in air.

Retained Heat Cooker: A relatively air-tight, wellinsulated box that uses captured heat to simmer a hot pot of food to completion.

Specific Consumption: The fuel used per unit of product produced, e.g., how much wood was used to cook a liter of beans.

Stratification: The levels of smoke and other pollutants that rise and can be more highly concentrated near the ceiling of a room.

Test Kitchen: A kitchen used for testing emissions in which the air exchanges are controlled to reduce the effect of ventilation on the measured levels of emissions.

Turn Down Ratio: The ratio between high and low power in a stove. The high firepower is divided by the low firepower.

Ventilation: The exchange of air from the outside to the inside of a building.

Water Boiling Test (WBT): A standardized test in which water is boiled and simmered. Fuel use and other parameters, including emissions, are measured. The WBT is designed to investigate the heat transfer and combustion characteristics of a stove under controlled operating procedures.

## Appendix B

## Testing Methods

## How were the tests performed and analyzed?

Many variables affect the performance of a cook stove. Whether the stove was cold or hot when started, the difference in performance when slowly simmering and rapidly boiling, and the skill of the operator all affect the test results.

A standard method for determining stove performance is the UCB 2003 Revised Water Boiling Test. This test has three phases:

1. Bringing 5 L of water to a boil at high power with the stove starting cold, or "cold start."
2. Bringing 5 L of water to a boil at high power with the stove starting hot, or "hot start."
3. Simmering 5 L of water for 45 minutes at low power ( $3^{\circ}$ to $6^{\circ} \mathrm{C}$ below full-boiling temperature).

The international standard 7 L stainless steel testing pot with no lid was used for each test for each stove except the alcohol stove.

Kiln-dried Douglas fir cut into sticks $1 \mathrm{~cm} \times 1.5 \mathrm{~cm}$ x 30 cm was used for fuel. The fan stoves were fueled by $5 \mathrm{~cm} \times 3 \mathrm{~cm} \times 1.5 \mathrm{~cm}$ pieces of the same wood. The fuel was carefully metered into the fire in an effort to operate each stove as effectively as possible.

The levels of emissions released during a Water Boiling Test (WBT) varied depending on how and where they are measured. Two approaches involve 1) collecting all the smoke under a hood and 2) monitoring the amount of smoke dispersed in the air of a test kitchen.

Emission testing provides information about how cleanly the stove changes fuel into useable heat. Emission testing can also shed light on how much carbon monoxide (CO), particulate matter (PM) and other pollutants are found in room air.

Three series of tests were performed on each stove:

1. WBT Series (three full WBTs per stove) - monitoring only fuel use, not emissions:
a. 5 L of water brought to a boil with stove at cold start.
b. 5 L of water brought to a boil with stove at hot start.
c. 5 L of water boiled again and then simmered for 45 minutes.
2. Test Kitchen Series (three per stove) - monitoring fuel use and emission concentration within an approximately $15 \mathrm{~m}^{3}$ kitchen:
a. 5 L of water brought to a boil with stove at cold start.
b. 5 L of water simmered for 30 minutes.
3. Emissions Hood Test Series (three full WBTs per stove) monitoring fuel use and collecting/recording total emissions released from each stove:
a. 5 L of water brought to a boil from a cold start.
b. 5 L of water brought to boil from a hot start.
c. 5 L of water simmered for 45 minutes.
(Due to technical problems, the PM data were not usable from two of the three WBTs performed under the emissions hood.)

## Emissions testing hood

The emissions testing hood at Aprovecho collects all of the smoke created by a fire and records the amount of pollutants created each second.

The emissions collection hood includes the following:

Hood. A $1 \mathrm{~m}^{2}$ bell with fire-resistant adjustable welder's fabric hanging from three sides. The hood may be raised or lowered depending on the size of the stove.

Exhaust System and Flow Measurements. The smoke is drawn up through the hood by using a large fan. The flow is adjusted so that the smoke is collected without inducing extra draft in the stove. Flow is measured with a manometer by pressure drop across a $1.5^{\prime \prime}$ diameter orifice and a type K thermocouple.

Gas Concentration Measurement. Concentrations of $\mathrm{CO}, \mathrm{CO}_{2}$ and hydrocarbons are measured after the orifice by an Enerac 3000E NDIR (infrared) stack meter.

Particulate Measurement. A sample of smoke is drawn from the exhaust, diluted and cooled with clean, dry air then metered using a Radiance Research Nephelometer with light-scattering analysis. The CO and $\mathrm{CO}_{2}$ are then measured again, using sensors provided by Tami Bond and Chris Roden of UIUC, to determine the level of dilution of the smoke sample.

Data Acquisition and Analysis. Analog signals from the sensors are read by a data acquisition board connected to a computer. Concentration data are displayed in real time on a computer monitor. Data are analyzed in conjunction with WBT data entered, using an Excel spreadsheet with a Visual Basic macro developed by Tami Bond and Nordica MacCarty. The concentration of each of the emission components times the mass flow through the hood can be integrated over time to calculate how much of each pollutant was produced during a given time period. When a standard WBT (representing a cooking task) is done under the hood, it is possible to determine how much wood is consumed and how much pollution is generated in performing the task.


## Test kitchen

The Aprovecho test kitchen is a building measuring $8 \times 10 \times 8 \mathrm{ft}$ designed to replicate common kitchens around the world. It has been calculated to have about three air exchanges per hour. The stove tester sitting inside the kitchen wears a forced-air respirator so that he or she can breathe fresh air from outside.


Emissions monitors consist of the following:

## AP Buck Personal Air Sampler measuring PM.

A common method for measuring PM is a pump and filter system that draws in air at a constant rate through a pre-weighed filter. The particles collect on the filter during the test. The filter is post weighed after the test on a very sensitive scale. The mass of the particles, factored by the rate of air flow through the filter and the amount of run time, gives the average concentration of PM entering the intake during the test. The flow rate of the pump is calibrated using an AP Buck bubble calibrator.

HOBO CO Loggers measuring CO. A common method for measuring concentrations of CO is the HOBO data logger. The HOBO uses an electrochemical cell, which puts off an electrical signal proportional to the concentration of CO in the air. The signal is recorded by an on-board data logger. The unit is launched and provides results on a personal computer, providing a moment-by-moment graph of the CO levels in the room.

Three HOBOs were used in the test kitchen: one logger 1.3 meters away from the stove, one at 1 meter off the floor and one 2.5 meters above the stove. CO tends to stratify, collecting near the ceiling. In this report, only the average readout of the HOBO 1.3 meters from the stove is reported.

In the test kitchen tests, 5 L of water were brought to a boil and then simmered for 30 minutes.

This report presents stove performance based on 10 measures of key importance. The final results were calculated as an average of the 18 total applicable test phases completed for each stove.

1. Time to Boil 5 L of water - Corrected to reflect a beginning temperature of $25^{\circ} \mathrm{C}$. Average of the following (11) tests:
a. One cold and three hot starts in the WBT.
b. Three cold starts in the test kitchen.
c. One cold and three hot starts in the emissions hood tests.
2. Fuel to Boil 1 L - Temperature-corrected specific consumption is a measure of fuel used per liter of boiling water produced, starting from a corrected temperature of $25^{\circ} \mathrm{C}$. Average of the following (11) tests:
a. One cold and three hot starts in the WBT.
b. Three cold starts in the test kitchen.
c. One cold and three hot starts in the emissions hood test.
3. Fuel to Simmer 1 L-Temperature-corrected specific consumption to produce 1 L of simmering water for 45 minutes, average from the following tests:
a. Three WBT.
b. Three emissions tests.
4. Fuel/Energy to Cook 5 L - found by adding the average fuel to boil 1 L to the average fuel to simmer 1 L for 45 minutes, a typical cooking
situation. This is multiplied by 5 L . When multiplied by the effective calorific value of the fuel used, a comparison of energy used is possible to compare stoves burning different fuels.
5. CO Emissions to Cook 5 L - Separate reporting from both methods of measuring.

## Emissions:

a. Emissions Hood - Monitoring the quantity of CO produced each second to find the grams of CO produced during each test phase. To find CO emissions to cook 5 L , the average grams of CO produced to boil 1 L in cold and hot starts is added to the CO produced to simmer 1 L for 45 minutes, averaged across three tests under the emissions hood.
b. Test Kitchen - The average of three tests reporting the average of the CO concentration recorded by a HOBO CO sensor at breathing level in the test kitchen for the duration of a cooking situation (boil 5 L and then simmer for 30 minutes).
6. Particulate Emissions to Cook 5 L - Separate reporting from both methods of measuring emissions:
a. Emissions Hood - Data from one WBT data monitoring the micrograms of PM emissions each second to find the specific milligrams of PM produced during each test phase. The average milligrams of PM produced to boil 1 L in cold and hot starts is added to the PM produced to simmer 1 L for 45 minutes under the emissions hood.
b. Test kitchen - Average of three tests of the average PM concentration recorded by an AP Buck Pump and Filter system at breathing level in the test kitchen during the duration of a cooking situation (boil 5 L and then simmer for 30 minutes).
7. Thermal Efficiency - Energy transferred into the water expressed as heating and vaporization divided by energy consumed from the wood. Average of all tests.
8. Firepower - Energy in the fuel consumed divided by the time of burning in seconds. Average of all tests.
9. $\mathrm{CO} / \mathrm{CO}_{2}$ Ratio - Grams of CO converted to moles divided by grams of $\mathrm{CO}_{2}$ converted to moles produced during each phase of testing. Average of all emissions hood tests.
10. Emission Factors - Mass of pollutant divided by mass of dry fuel consumed during the test phase. Average of all emissions hood tests.
11. Turn Down Ratio - High-power firepower divided by Low-power firepower.

The following pages contain the full testing data and information on variation between tests.

## Appendix C

## Testing Data

Calculations and Theory for the UCB 2003 Revised Water Boiling Test<br>\section*{Variables that are directly measured}<br>$\mathrm{f}_{\text {hi }} \quad$ Weight of fuel before test (grams)<br>$\mathrm{P}_{\mathrm{hi}} \quad$ Weight of pot with water before test (grams)<br>$\mathrm{T}_{\text {hi }} \quad$ Water temperature before test $\left({ }^{\circ} \mathrm{C}\right)$<br>$\mathrm{t}_{\mathrm{hi}} \quad$ Time at start of test (min)<br>$\mathrm{f}_{\mathrm{hf}} \quad$ Weight of wood after test (grams)<br>$c_{h} \quad$ Weight of charcoal and container after test (grams)<br>$\mathrm{P}_{\mathrm{hf}} \quad$ Weight of pot with water after test (grams)<br>$\mathrm{T}_{\mathrm{hf}} \quad$ Water temperature after test $\left({ }^{\circ} \mathrm{C}\right)$<br>$\mathrm{t}_{\mathrm{hf}} \quad$ Time at end of test (min)

## Variables that are calculated

$\mathrm{f}_{\mathrm{hm}} \quad$ Wood consumed, moist (grams)

$$
\mathrm{f}_{\mathrm{hm}}=\mathrm{f}_{\mathrm{hf}}-\mathrm{f}_{\mathrm{hi}}
$$

${ }^{\bullet} c_{h} \quad$ Net change in char during test phase (grams) $\cdot{ }^{\circ} c_{h}=c_{h}-k$
$\mathrm{f}_{\mathrm{hd}} \quad$ Equivalent dry wood consumed (grams)
$\mathrm{f}_{\mathrm{hd}}=\mathrm{f}_{\mathrm{hm}}^{*}\left(1-\left(1.12^{*} \mathrm{~m}\right)\right)-1.5^{*} \Delta \mathrm{c}_{\mathrm{h}}$
$\mathrm{w}_{\mathrm{hv}} \quad$ Water vaporized (grams)
$\mathrm{w}_{\mathrm{hv}}=\mathrm{P}_{\mathrm{hi}}-\mathrm{P}_{\mathrm{hf}}$
$\mathrm{w}_{\mathrm{hr}} \quad$ Water remaining at end of test (grams)
$\mathrm{w}_{\mathrm{hr}}=\mathrm{P}_{\mathrm{hf}}-\mathrm{P}$

- $\mathrm{t}_{\mathrm{h}} \quad$ Duration of phase (min)
$h_{h} \quad$ Thermal efficiency
$\bullet \mathrm{t}_{\mathrm{h}}=\mathrm{t}_{\mathrm{hf}}-\mathrm{t}_{\mathrm{hi}}$
$\mathrm{h}=\frac{4.186 *\left(\mathrm{P}_{\mathrm{hi}}-\mathrm{P}\right) *\left(\mathrm{~T}_{\mathrm{hf}}-\mathrm{T}_{\mathrm{hh}}\right)+2260 *\left(\mathrm{~W}_{\mathrm{hv}}\right)}{\mathrm{f}_{\mathrm{hd}} * \mathrm{LHV}}$
$\mathrm{r}_{\mathrm{hb}} \quad$ Burning rate (grams/min)
$\mathrm{r}_{\mathrm{hb}}=\frac{\mathrm{f}_{\mathrm{hd}}}{\mathrm{t}_{\mathrm{hi}}-\mathrm{t}_{\mathrm{hf}}}$
$\mathrm{SC}_{\mathrm{h}} \quad$ Specific fuel consumption
(grams wood/grams water)
$S C_{h}=\frac{f_{h d}}{P_{h f}-P}$
$\mathrm{SC}^{\mathrm{T}} \quad$ Temp-corrected specific consumption (grams wood/grams water)
$\mathrm{SC}_{\mathrm{h}}^{\mathrm{T}}=\frac{\mathrm{f}_{\mathrm{hd}}}{\mathrm{P}_{\mathrm{hf}}-\mathrm{P}} * \frac{75}{\mathrm{~T}_{\mathrm{hf}}-\mathrm{T}_{\mathrm{hi}}}$
$\mathrm{FP}_{\mathrm{h}} \quad$ Firepower (W)
$\mathrm{FP}_{\mathrm{h}}=\frac{\mathrm{f}_{\mathrm{hd}} * L H V}{60 *\left(\mathrm{t}_{\mathrm{hi}}-\mathrm{t}_{\mathrm{hf}}\right)}$
TDR Turn down ratio
$\mathrm{TDR}=\frac{\mathrm{FP}_{\mathrm{h}}}{\mathrm{FP}_{\mathrm{s}}}$


## Explanations of Calculations

$\mathrm{f}_{\mathrm{cm}}$ - Wood consumed (moist): This is the mass of wood that was used to bring the water to a boil found by taking the difference of the pre-weighed bundle of wood and the wood remaining at the end of the test phase:

$$
\mathrm{f}_{\mathrm{cm}}=\mathrm{f}_{\mathrm{cf}}-\mathrm{f}_{\mathrm{ci}}
$$

${ }^{-} \mathrm{c}_{\mathrm{c}}$ - Net change in char during test phase: This is the mass of char created during the test found by removing the char from the stove at the end of the test phase. Because it is very hot, the char will be placed in an empty pre-weighed container of mass k (to be supplied by testers) and weighing the char with the container, then subtracting the two masses.

$$
\bullet \boldsymbol{c}_{\mathrm{c}}=\mathrm{c}_{\mathrm{c}}-\mathrm{k}
$$

$\mathrm{f}_{\mathrm{cd}}$ - Equivalent dry wood consumed: This is a calculation that adjusts the amount of wood that was burned in order to account for two factors: (1) the energy that was needed to remove the moisture in the wood and (2) the amount of char remaining unburned. The calculation is done in the following way:

$$
f_{c d}=f_{c m} *(1-(1.12 * m))-1.5 * \Delta c_{c}
$$

The factor of $1-(1.12 * \mathrm{~m})$ adjusts the mass of wood burned by the amount of wood required to heat and evaporate $\mathrm{m} * \mathrm{f}_{\mathrm{cm}}$ grams of water. It takes roughly $2,260 \mathrm{~kJ}$ to evaporate a kilogram of water, which is roughly $12 \%$ of the calorific value of dry wood. Thus if wood consists of $\mathrm{m} \%$ moisture, the mass of wood that can effectively heat the pot of water is reduced by roughly $1-(1.12 * \mathrm{~m})$ because the water must be boiled away (see Baldwin, 1986 for further discussion).

The factor of $1.5 * \Delta c_{c}$ accounts for the wood converted into unburned char. Char has roughly $150 \%$ the calorific content of wood, thus the amount of wood heating the pot of water should be adjusted by $1.5 * \Delta \mathrm{c}_{\mathrm{c}}$ to account for the remaining char. Note, in the simmer phase it is possible that there will be a net loss in the amount of char before and after the test, in which case $\cdot \mathrm{c}$ is negative and the equivalent dry wood increases rather than decreases.
$\mathbf{w}_{\mathrm{cv}}$ - Water vaporized: This is a measure of the amount of water lost through evaporation during the test. It is calculated by subtracting the final weight of pot and water from the initial weight of pot and water.

$$
\mathrm{w}_{\mathrm{cv}}=\mathrm{P}_{\mathrm{ci}}-\mathrm{P}_{\mathrm{cf}}
$$

$\mathbf{w}_{\mathrm{cr}}-$ Water remaining at end of test: This is a measure of the amount of water heated to boiling. It is calculated by subtracting the weight of the pot from the final weight of the pot and water.

$$
\mathrm{w}_{\mathrm{cr}}=\mathrm{P}_{\mathrm{cf}}-\mathrm{P}
$$

${ }^{\bullet} \boldsymbol{t}_{\mathrm{c}}$ - Duration of phase: This is simply the time taken to perform the test. It is a simple clock difference:

$$
{ }^{\bullet t_{c}}=\mathrm{t}_{\mathrm{cf}}-\mathrm{t}_{\mathrm{ci}}
$$

$\mathbf{h}_{\mathbf{c}}$ - Thermal efficiency: This is a ratio of the work done by heating and evaporating water to the energy consumed by burning wood. It is calculated in the following way.

$$
\mathrm{h}_{\mathrm{c}}=\frac{4.186 *\left(\mathrm{P}_{\mathrm{ci}}-\mathrm{P}\right) *\left(\mathrm{~T}_{\mathrm{cf}}-\mathrm{T}_{\mathrm{ci}}\right)+2260 *\left(\mathrm{w}_{\mathrm{cv}}\right)}{\mathrm{f}_{\mathrm{cd}} * \mathrm{LHV}}
$$

In this calculation, the work done by heating water is determined by adding two quantities: (1) the product of the mass of water in the pot, $\left(\mathrm{P}_{\mathrm{ci}}-\mathrm{P}\right)$, the specific heat of water (4.186 $\mathrm{J} / \mathrm{g}^{\circ} \mathrm{C}$ ), and the change in water temperature ( $\mathrm{T}_{\mathrm{cf}}-\mathrm{T}_{\mathrm{c}}$ ) and (2) the product of the amount of water evaporated from the pot and the latent heat of evaporation of water $(2,260 \mathrm{~J} / \mathrm{g})$. The denominator (bottom of the ratio) is determined by taking the product of the drywood equivalent consumed during this phase of the test and the lower heat value (LHV).
$\mathbf{r}_{\mathrm{cb}}$ - Burning rate: This is a measure of the rate of wood consumption while bringing water to a boil. It is calculated by dividing the equivalent dry wood consumed by the time of the test.

$$
r_{c b}=\frac{f_{c d}}{t_{\mathrm{ci}}-t_{\mathrm{cf}}}
$$

$\mathrm{SC}_{\mathrm{c}}$ - Specific fuel consumption: Specific consumption can be defined for any number of cooking tasks and should be considered "the fuelwood required to produce a unit output" whether the output is boiled water, cooked beans, or loaves of bread. In the case of the coldstart high-power WBT, it is a measure of the amount of wood required to produce one liter (or kilo) of boiling water starting with cold stove. It is calculated in this way:

$$
\mathrm{SC}_{\mathrm{c}}=\frac{\mathrm{f}_{\mathrm{cd}}}{\mathrm{P}_{\mathrm{cf}}-\mathrm{P}}
$$

$S C^{\mathrm{T}}{ }_{\mathrm{c}}$ - Temperature corrected specific fuel consumption: This corrects specific consumption to account for differences in initial water temperatures. This facilitates comparison of stoves tested on different days or in different environmental conditions. The correction is a simple factor that "normalizes" the temperature change observed in test conditions to a "standard" temperature change of $75^{\circ} \mathrm{C}$ (from 25 to 100). It is calculated in the following way.

$$
S C^{\top}{ }_{c}=\frac{f_{c d}}{P_{c f}-P} * \frac{75}{T_{c f}-T_{c i}}
$$

$\mathrm{FP}_{\mathrm{c}}$ - Firepower: This is a ratio of the wood energy consumed by the stove per unit time. It tells the average power output of the stove (in watts) during the high-power test.

$$
\mathrm{FP}_{\mathrm{c}}=\frac{\mathrm{f}_{\mathrm{cd}} * \mathrm{LHV}}{60 *\left(\mathrm{t}_{\mathrm{ci}}-\mathrm{t}_{\mathrm{cf}}\right)}
$$

Note, by using $\mathrm{f}_{\mathrm{cd}}$ in this calculation, we have accounted for both the remaining char and the wood moisture content.

HOOD Results
3 stone Ghana 20 LCan Mud/
fire wood Rocket Sawdust VITA

| 1. HIGH POWER TEST (COLD START) | units |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Time to boil Pot \# 1 | min | 28 | 25 | 22 | 20 | 16 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 24.08 | 21.03 | 17.13 | 24.15 | 25.18 |
| Thermal efficiency | \% | 19\% | 24\% | 37\% | 28\% | 29\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 136.02 | 106.38 | 76.70 | 94.28 | 83.71 |
| Temp-corrected specific consumption | $\mathrm{g} / \mathrm{L}$ | 118.44 | 92.13 | 68.06 | 82.02 | 72.91 |
| Firepower | watts | 7,761 | 6,774 | 5,532 | 7,801 | 8,129 |
| Equivalent dry wood consumed | g | 511 | 387.4 | 395.3 | 412.5 | 269.6 |
| 2. HIGH POWER TEST (HOT START) | units |  |  |  |  |  |
| Time to boil Pot \# 1 | min | 30 | 22 | 23 | 16 | 15 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 25.61 | 19.32 | 18.03 | 24.86 | 24.66 |
| Thermal efficiency | \% | 20\% | 27\% | 31\% | 31\% | 31\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 136.87 | 86.67 | 85.99 | 83.30 | 74.97 |
| Temp-corrected specific consumption | $\mathrm{g} / \mathrm{L}$ | 121.92 | 76.54 | 76.18 | 72.58 | 67.82 |
| Firepower | watts | 8,243 | 6,207 | 5,809 | 8,004 | 7,944 |
| Equivalent dry wood consumed | g | 567.6 | 310.0 | 416.0 | 390.9 | 273.6 |
| 3. LOW POWER (SIMMER) | units |  |  |  |  |  |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 9.49 | 9.91 | 6.68 | 6.28 | 7.15 |
| Thermal efficiency | \% | 26\% | 23\% | 26\% | 44\% | 34\% |
| Specific fuel consumption | g/L | 103.38 | 114.89 | 74.41 | 81.28 | 67.52 |
| Firepower | watts | 3,130 | 3,298 | 2,235 | 2,078 | 2,385 |
| Turn down ratio |  | 2.77 | 1.99 | 2.64 | 3.92 | 3.85 |
| Equivalent dry wood consumed | g | 419.4 | 279.1 | 192.8 | 221.5 | 208.8 |
| 4. ENERGY \& MOISTURE CONTENT OF FUEL | units |  |  |  |  |  |
| Net calorific value (dry) | kJ/kg | 19,260 | 19,260 | 19,260 | 19,260 | 19,260 |
| Moisture content | \% | 11\% | 11\% | 11\% | 11\% | 11\% |
| Effective calorific fuel value | kJ/kg | 17,302 | 17,332 | 17,332 | 17,334 | 17,281 |
| 5. COLD START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 23.8 | 24 | 22.7 | 18.1 | 14.1 |
| Energy consumption rate | kJ/min | 408 | 327 | 261 | 343 | 393 |
| Temp-Corrected specific energy consumption | kJ/L | 2,024 | 1,619 | 1,262 | 1,293 | 1,122 |
| Total energy consumed | kJ | 11,282 | 8,968 | 6,853 | 7,152 | 6,213 |
| 6. HOT START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 29.6 | 19.7 | 21.9 | 13.9 | 13.9 |
| Energy consumption rate | kJ/min | 431 | 318 | 296 | 428 | 412 |
| Temp-Corrected specific energy consumption | kJ/L | 2,160 | 1,298 | 1,337 | 1,223 | 1,149 |
| Total energy consumed | kJ | 11,766 | 7,173 | 7,220 | 6,784 | 6,314 |
| 7. SIMMER ADDITIONAL MEASURES | units |  |  |  |  |  |
| Energy consumption rate | kJ/min | 161 | 143 | 111 | 114 | 107 |
| Temp-Corrected specific energy consumption | kJ/L | 1,807 | 1,580 | 1,216 | 1,364 | 1,175 |
| Total energy consumed | kJ | 7,625 | 6,455 | 4,977 | 5,137 | 4,813 |

HOOD Results

| OOD Resuits |  | Justa | Uganda 2-pot |  | Onil | EcoStove |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1. HIGH POWER TEST (COLD START) | units |  |  |  |  |  |
| Time to boil Pot \# 1 | min | 52 | 20 | 42 | 35 | 53 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 25.41 | 20.38 | 25.57 | 33.58 | 29.87 |
| Thermal efficiency | \% | 17\% | 40\% | 20\% | 18\% | 13\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 150.86 | 60.84 | 123.31 | 139.95 | 296.04 |
| Temp-corrected specific consumption | g/L | 130.11 | 52.41 | 108.29 | 118.98 | 260.29 |
| Firepower | watts | 8,203 | 6,577 | 8,212 | 10,829 | 8,998 |
| Equivalent dry wood consumed | g | 884.5 | 265.6 | 709.2 | 743.3 | 1074.4 |
| 2. HIGH POWER TEST (HOT START) | units |  |  |  |  |  |
| Time to boil Pot \# 1 | min | 39 | 15 | 33 | 28 | 34 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 29.62 | 23.50 | 26.27 | 32.53 | 31.84 |
| Thermal efficiency | \% | 21\% | 45\% | 24\% | 22\% | 16\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 151.80 | 58.63 | 129.78 | 131.72 | 234.21 |
| Temp-corrected specific consumption | g/L | 134.75 | 52.29 | 114.79 | 114.77 | 208.16 |
| Firepower | watts | 8,685 | 7,580 | 8,439 | 10,489 | 9,626 |
| Equivalent dry wood consumed | g | 703.2 | 231.3 | 529.6 | 551.8 | 735.6 |
| 3. LOW POWER (SIMMER) | units |  |  |  |  |  |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 12.70 | 7.71 | 12.96 | 14.50 | 14.67 |
| Thermal efficiency | \% | 14\% | 33\% | 14\% | 13\% | 16\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 140.90 | 91.71 | 143.93 | 160.32 | 168.50 |
| Firepower | watts | 4,180 | 2,550 | 4,253 | 4,796 | 4,531 |
| Turn down ratio |  | 2.03 | 3.11 | 1.99 | 2.24 | 2.04 |
| Equivalent dry wood consumed | g | 392.5 | 214.9 | 401.3 | 409 | 448.8 |
| 4. ENERGY \& MOISTURE CONTENT OF FUEL | units |  |  |  |  |  |
| Net calorific value (dry) | kJ/kg | 19,260 | 19,260 | 19,260 | 19,260 | 19,260 |
| Moisture content | \% | 11\% | 11\% | 11\% | 11\% | 11\% |
| Effective calorific fuel value | kJ/kg | 17,384 | 17,384 | 17,384 | 17,345 | 17,284 |
| 5. COLD START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 54.5 | 18.7 | 40.1 | 30.4 | 47.8 |
| Energy consumption rate | kJ/min | 363 | 310 | 397 | 541 | 521 |
| Temp-Corrected specific energy consumption | kJ/L | 2,437 | 843 | 1,869 | 1,942 | 5,338 |
| Total energy consumed | kJ | 23,080 | 6,914 | 18,530 | 19,356 | 28,032 |
| 6. HOT START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 38.9 | 13.6 | 29.6 | 25.6 | 29.5 |
| Energy consumption rate | $\mathrm{kJ} / \mathrm{min}$ | 420 | 377 | 413 | 489 | 571 |
| Temp-Corrected specific energy consumption | kJ/L | 2,006 | 759 | 1,463 | 1,474 | 3,642 |
| Total energy consumed | kJ | 18,322 | 6,021 | 13,811 | 14,369 | 19,245 |
| 7. SIMMER ADDITIONAL MEASURES | units |  |  |  |  |  |
| Energy consumption rate | $\mathrm{kJ} / \mathrm{min}$ | 228 | 124 | 233 | 237 | 261 |
| Temp-Corrected specific energy consumption | kJ/L | 2,493 | 1,475 | 2,599 | 2,592 | 2,989 |
| Total energy consumed | kJ | 10,239 | 5,609 | 10,490 | 10,648 | 11,734 |

HOOD Results

|  |  | Flame Fan | Gas <br> Fan | Charcoal | Charcoal | Propane |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1. HIGH POWER TEST (COLD START) | units |  |  |  |  |  |
| Time to boil Pot \# 1 | min | 23 | 29 | 38 | 37 | 32 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 12.68 | 8.2 | 11.10 | 10.97 | 2.43 |
| Thermal efficiency | \% | 42\% | 45\% | 17\% | 18\% | 69\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 59.37 | 53.46 | 88.55 | 81.28 | 14.94 |
| Temp-corrected specific consumption | g/L | 49.85 | 47.22 | 78.31 | 70.65 | 12.70 |
| Firepower | watts | 4,093 | 2,656 | 5,859 | 5,790 | 1,946 |
| Equivalent dry wood consumed | g | 265.9 | 206.9 | 253.2 | 256.8 | 66 |
| 2. HIGH POWER TEST (HOT START) | units |  |  |  |  |  |
| Time to boil Pot \# 1 | min | 23 | 29 | 47 | 29 | 30 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 12.43 | 8.5 | 10.31 | 12.76 | 2.40 |
| Thermal efficiency | \% | 42\% | 46\% | 18\% | 19\% | 66\% |
| Specific fuel consumption | $\mathrm{g} / \mathrm{L}$ | 58.65 | 51.05 | 93.34 | 75.02 | 14.86 |
| Temp-corrected specific consumption | g/L | 49.66 | 46.60 | 83.97 | 65.95 | 12.94 |
| Firepower | watts | 4,003 | 2,761 | 5,443 | 6,735 | 1,915 |
| Equivalent dry wood consumed | g | 278.8 | 0.0 | 272.9 | 230.9 | 67 |
| 3. LOW POWER (SIMMER) | units |  |  |  |  |  |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 6.21 | 4.20 | 4.84 | 5.89 | 1.32 |
| Thermal efficiency | \% | 42\% | 46\% | 27\% | 34\% | 61\% |
| Specific fuel consumption | g/L | 75.4 | 44.85 | 53.64 | 70.54 | 15.00 |
| Firepower | watts | 2,059 | 1,400 | 2,586 | 3,174 | 1,072 |
| Turn down ratio |  | 1.88 | 1.97 | 2.10 | 2.82 | 1.89 |
| Equivalent dry wood consumed | g | 276.5 | 0.0 | 157.9 | 137.6 | 41.3 |
| 4. ENERGY \& MOISTURE CONTENT OF FUEL | units |  |  |  |  |  |
| Net calorific value (dry) | kJ/kg | 19,260 | 19,260 | 31,680 | 31,680 | 47,490 |
| Moisture content | \% | 12\% | 12\% | 6\% | 6\% | 0\% |
| Effective calorific fuel value | kJ/kg | 17,258 | 17,196 | 29,983 | 29,983 | 47,490 |
| 5. COLD START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 19.6 | 23.7 | 34.5 | 34.0 | 20.9 |
| Energy consumption rate | kJ/min | 200 | 124 | 291 | 289 | 136 |
| Temp-Corrected specific energy consumption | kJ/L | 816 | 755 | 2,081 | 2,035 | 589 |
| Total energy consumed | kJ | 4,587 | 3,558 | 11,370 | 11,540 | 1,606 |
| 6. HOT START ADDITIONAL MEASURES | units |  |  |  |  |  |
| Temp-Corrected time to boil | min | 19.4 | 23.7 | 42.7 | 22.7 | 25.0 |
| Energy consumption rate | kJ/min | 214 | 124 | 297 | 392 | 114 |
| Temp-Corrected specific energy consumption | kJ/L | 856 | 755 | 2,321 | 1,821 | 593 |
| Total energy consumed | kJ | 4,809 | 3,558 | 12,191 | 10,357 | 1,631 |
| 7. SIMMER ADDITIONAL MEASURES | units |  |  |  |  |  |
| Energy consumption rate | kJ/min | 106 | 124 | 161 | 141 | 65 |
| Temp-Corrected specific energy consumption | kJ/L | 1,266 | 1,132 | 1,759 | 1,674 | 743 |
| Total energy consumed | kJ | 4,773 | 5,337 | 7,259 | 6,332 | 2,949 |

HOOD Results

| Alcohol |  |  |
| :---: | :---: | :---: |
| Clean | Kero- | Solar |
| Cook | sene |  |


| 1. HIGH POWER TEST (COLD START) | units |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Time to boil Pot \# 1 | min | 38 | 46 | 76 |
| Burning rate | $\mathrm{g} / \mathrm{min}$ | 4.33 | 2.73 |  |
| Thermal efficiency | \% | 66\% | 52\% | 28\% |
| Specific fuel consumption | g/L | 34.44 | 25.82 |  |
| Temp-corrected specific consumption | g/L | 28.68 | 22.25 |  |
| Firepower | watts | 1,544 | 1,859 | 2,386 |
| Equivalent dry wood consumed | g | 165.0 | 114.0 |  |
| 2. HIGH POWER TEST (HOT START) | units |  |  |  |
| Time to boil Pot \# 1 | min | 38 | 51 | 77 |
| Burning rate | $\mathrm{g} /$ min | 4.33 | 2.56 |  |
| Thermal efficiency | \% | 66\% | 51\% | 23\% |
| Specific fuel consumption | g/L | 34.44 | 27.67 |  |
| Temp-corrected specific consumption | g/L | 28.68 | 23.75 |  |
| Firepower | watts | 1,544 | 1,859 | 2,386 |
| Equivalent dry wood consumed | g | 165.0 | 113.0 |  |
| 3. LOW POWER (SIMMER) | units |  |  |  |
| Burning rate | $\mathrm{g} /$ min | 3.09 | 2.40 |  |
| Thermal efficiency | \% | 59\% | 40\% |  |
| Specific fuel consumption | g/L | 34.64 | 26.37 |  |
| Firepower | watts | 1,100 | 1,799 | 1,383 |
| Turn down ratio |  | 1.40 | 110 |  |
| Equivalent dry wood consumed | g | 139.0 | 96.0 |  |
| 4. ENERGY \& MOISTURE CONTENT OF FUEL | units |  |  |  |
| Net calorific value (dry) | kJ/kg | 21,370 | 43,500 |  |
| Moisture content | \% | 0\% |  |  |
| Effective calorific fuel value | kJ/kg | 21,370 | 43,500 |  |
| 5. COLD START ADDITIONAL MEASURES | units |  |  |  |
| Temp-Corrected time to boil | min | 31.6 | 40.7 | 69.9 |
| Energy consumption rate | $\mathrm{kJ} / \mathrm{min}$ | 93 | 104 |  |
| Temp-Corrected specific energy consumption | kJ/L | 613 | 916 |  |
| Total energy consumed | kJ | 3,526 | 4,959 |  |
| 6. HOT START ADDITIONAL MEASURES | units |  |  |  |
| Temp-Corrected time to boil | min | 31.6 | 43.1 | 70.0 |
| Energy consumption rate | $\mathrm{kJ} / \mathrm{min}$ | 93 | 105 |  |
| Temp-Corrected specific energy consumption | kJ/L | 613 | 911 |  |
| Total energy consumed | kJ | 3,526 | 5,307 |  |
| 7. SIMMER ADDITIONAL MEASURES | units |  |  |  |
| Energy consumption rate | $\mathrm{kJ} / \mathrm{min}$ | 66 | 93 |  |
| Temp-Corrected specific energy consumption | $\mathrm{kJ} / \mathrm{L}$ | 740 | 1,011 |  |
| Total energy consumed | kJ | 2,970 | 4,176 |  |

## HOOD Results

|  |  | 3 stone fire | Ghana <br> Wood | 20 L Can Rocket | Mud/ Sawdust | VITA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Totals | grams <br> grams <br> grams (prop) mg <br> degrees C |  |  |  |  |  |
| CO |  | 26.50 | 29.25 | 5.27 | 18.01 | 24.59 |
| CO 2 |  | 818 | 737 | 836 | 628 | 515 |
| HC (propane) |  | 1.7405 | 2.0569 | 0.5386 | 1.5064 | 1.5081 |
| appx PM |  | 1793 | 5294 | 997 | 1847 | 1887 |
| CO/CO2 ratio |  | 0.0572 | 0.0648 | 0.0101 | 0.0478 | 0.0752 |
| Flame temp |  | 485 | 159 | 226 | 155 | 282 |
|  |  |  |  |  |  |  |
| Totals | grams <br> grams <br> grams (prop) <br> mg <br> degrees C |  |  |  |  |  |
| CO |  | 28.91 | 20.22 | 6.50 | 16.39 | 26.19 |
| CO2 |  | 934 | 669 | 784 | 523 | 523 |
| HC (propane) |  | 2.3698 | 2.0739 | 0.7755 | 1.5788 | 1.9752 |
| appx PM gr |  | 2004 | 3751 | 594 | 2255 | 2642 |
| CO/CO2 ratio |  | 0.0533 | 0.0479 | 0.0137 | 0.0523 | 0.0804 |
| Flame temp |  | 300 | 155 | 177 | 164 | 301 |
|  |  |  |  |  |  |  |
| Totals | grams <br> grams <br> grams (prop) grams |  |  |  |  |  |
| CO |  | 31.43 | 24.71 | 9.95 | 29.90 | 21.88 |
| CO 2 |  | 815 | 682 | 830 | 735 | 661 |
| HC (propane) |  | 3.6547 | 3.8498 | 2.6255 | 4.1328 | 3.4533 |
| appx PM gr |  | 281 | 948 | MISS | 663 | 421 |
| CO/CO2 ratio |  | 0.0694 | 0.0594 | 0.0188 | 0.0645 | 0.0525 |
| Flame temp |  | 207 | 77 | 121 | 62 | 70 |

## (Corrected for water temp and Moisture)

Correction Factor 0.1408

| CO | $\mathrm{g} / \mathrm{L}$ <br> g/L <br> g/L <br> mg/L | 3.6351 | 3.7049 | 0.8044 | 2.3990 | 3.3932 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO 2 |  | 112.0006 | 94.4812 | 126.7232 | 82.3402 | 71.3697 |
| HC (propane) |  | 0.2385 | 0.2604 | 0.0800 | 0.1990 | 0.2090 |
| appx PM mg |  | 238.2039 | 652.9593 | 162.3765 | 265.2436 | 283.4517 |


| Correction Factor | 0.1446 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | $\mathrm{g} / \mathrm{L}$ | 4.0321 | 2.4414 | 0.9921 | 2.2284 | 3.6219 |
| CO 2 | $\mathrm{g} / \mathrm{L}$ | 130.2379 | 82.4174 | 119.2356 | 69.2491 | 72.3851 |
| HC (propane) | g/L | 0.3298 | 0.2511 | 0.1155 | 0.2118 | 0.2757 |
| appx PM mg | mg/L | 277.0760 | 414.9405 | 95.4476 | 324.7657 | 392.1768 |


| Correction Factor | 0.2274 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | $\mathrm{g} / \mathrm{L}$ | 7.3046 | 7.0095 | 2.1654 | 7.3922 | 5.0520 |
| CO 2 | $\mathrm{g} / \mathrm{L}$ | 185.1880 | 185.2308 | 180.5102 | 178.7203 | 154.5302 |
| HC (propane) | $\mathrm{g} / \mathrm{L}$ | 0.8474 | 1.0770 | 0.5715 | 0.9864 | 0.8090 |
| appx PM mg | mg/L | 214.9977 | 323.4849 | 128.9121 | 175.4471 | 92.2347 |

HOOD Results

|  |  | Justa | Uganda 2pot | Prototype | Onil | Eco-stove |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Totals | grams grams grams (prop) mg <br> degrees C |  |  |  |  |  |
| CO |  | 26.93 | 13.03 | 16.47 | 32.89 | 52.83 |
| CO 2 |  | 1983 | 372 | 1700 | 1730 | 2382 |
| HC (propane) |  | 5.1406 | 1.4423 | 2.6841 | 2.4597 | 4.1091 |
| appx PM |  | 983 | 662 | 903 | 1437 | 5535 |
| CO/CO2 ratio |  | 0.0200 | 0.0539 | 0.0147 | 0.0286 | 0.0349 |
| Flame temp |  | 11 | 12 | 12 | 396 | 13 |
|  |  |  |  |  |  |  |
| Totals | grams grams grams (prop) mg <br> degrees C |  |  |  |  |  |
| CO |  | 17.40 | 12.46 | 13.18 | 17.19 | 27.41 |
| CO2 |  | 1726 | 369 | 1549 | 1361 | 1442 |
| HC (propane) |  | 5.1970 | 2.0540 | 3.1776 | 3.1035 | 3.1484 |
| appx PM gr |  | 933 | 813 | 835 | 1483 | 4343 |
| CO/CO2 ratio |  | 0.0148 | 0.0529 | 0.0146 | 0.0193 | 0.0302 |
| Flame temp |  | 13 | 13 | 13 | 442 | 13 |
|  |  |  |  |  |  |  |
| Totals | grams grams grams (prop) grams |  |  |  |  |  |
| CO |  | 12.80 | 12.87 | 11.96 | 18.10 | 14.06 |
| CO2 |  | 1254 | 654 | 1010 | 1141 | 1077 |
| HC (propane) |  | 5.8437 | 4.8623 | 4.1460 | 4.2688 | 4.3466 |
| appx PM gr |  | 340 | 264 | 446 | 671 | 1044 |
| CO/CO2 ratio |  | 0.0168 | 0.0304 | 0.0205 | 0.0247 | 0.0201 |
| Flame temp |  | 13 | 14 | 12 | 277 | 13 |

(Corrected for water temp and moisture)

| Correction Factor | 0.1408 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | $\mathrm{g} / \mathrm{L}$ | 2.3540 | 1.1894 | 1.2863 | 2.6733 | 8.1161 |
| CO2 | $\mathrm{g} / \mathrm{L}$ | 172.6200 | 33.9872 | 132.9404 | 139.3402 | 365.8552 |
| HC (propane) | $g / L$ | 0.4454 | 0.1327 | 0.2106 | 0.1952 | 0.6313 |
| appx PM mg | mg/L | 83.1819 | 62.6387 | 71.8650 | 111.7469 | 851.8434 |


| Correction Factor | 0.1446 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | g/L <br> g/L <br> g/L <br> mg/L | 1.5561 | 1.1529 | 1.0826 | 1.3988 | 4.0083 |
| CO2 |  | 153.5459 | 34.0263 | 127.0328 | 109.4260 | 211.2221 |
| HC (propane) |  | 0.4603 | 0.1871 | 0.2610 | 0.2401 | 0.4607 |
| appx PM mg |  | 78.5652 | 72.2213 | 69.3188 | 112.0330 | 642.2649 |
| Correction Factor | 0.2274 |  |  |  |  |  |
| CO | $\begin{aligned} & g / L \\ & g / L \end{aligned}$ | 2.8703 | 3.2840 | 2.7024 | 4.2671 | 3.5458 |
| CO 2 |  | 280.5839 | 166.6359 | 224.6289 | 269.0215 | 266.7382 |
| HC (propane) | $\mathrm{g} / \mathrm{L}$ | 1.3105 | 1.2417 | 0.9257 | 1.0003 | 1.0712 |
| appx PM mg | mg/L | 77.6056 | 68.1982 | 105.3058 | 156.6180 | 273.3383 |

HOOD Results
$\begin{array}{ll}\text { Gyapa } & \\ \text { Char- } & \text { Pro- }\end{array}$

| Totals | grams <br> grams <br> grams (prop) <br> mg <br> degrees C |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO |  | 6.49 | 3.64 | 73.97 | 105.09 | 0.64 |
| CO2 |  | 510 | 525 | 524 | 665 | 281 |
| HC (propane) |  | 1.2411 | 2.3581 | 8.0845 | 7.7670 | 0.9349 |
| appx PM |  | 6 | 14 | 1026 | 1155 | 2 |
| CO/CO2 ratio |  | 0.0201 | 0.0109 | 0.2219 | 0.2522 | 0.0043 |
| Flame temp |  | 228 | 366 | 181 | 312 | 13 |
|  |  |  |  |  |  |  |
| Totals | grams <br> grams <br> grams (prop) <br> mg <br> degrees C |  |  |  |  |  |
| CO |  | 5.14 | 3.64 | 75.57 | 85.79 | 0.50 |
| CO2 |  | 504 | 525 | 577 | 630 | 323 |
| HC (propane) |  | 1.5259 | 2.3581 | 10.8352 | 18.7060 | 1.5905 |
| appx PM gr |  | 48 | 14 | 1149 | 1656 | 1 |
| CO/CO2 ratio |  | 0.0182 | 0.0109 | 0.2279 | 0.1579 | 0.0025 |
| Flame temp |  | 251 | 366 | 214 | 360 | 13 |
|  |  |  |  |  |  |  |
| Totals | grams <br> grams <br> grams (prop) <br> grams |  |  |  |  |  |
| CO |  | 4.52 | 5.46 | 43.41 | 56.89 | 0.02 |
| CO2 |  | 664 | 788 | 359 | 522 | 341 |
| HC (propane) |  | 3.5576 | 3.5372 | 7.8860 | 10.9529 | 4.3930 |
| appx PM gr |  | 25 | 22 | 162 | 169 | 2 |
| CO/CO2 ratio |  | 0.0117 | 0.0109 | 0.2251 | 0.1726 | 0.0001 |
| Flame temp |  | 180 |  | 187 | 202 | 11 |

(Corrected for water temp and moisture)

| Correction Factor | 0.1408 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | g/L | 0.9190 | 0.5488 | 12.7035 | 15.5781 | 0.1164 |
| CO2 | g/L | 71.9071 | 79.2075 | 90.0783 | 98.5886 | 50.3029 |
| HC (propane) | g/L | 0.1759 | 0.3557 | 1.3648 | 1.1513 | 0.1661 |
| appx PM mg | mg/L | 0.8663 | 2.1627 | 50.2483 | 102.5396 | 0.2854 |


| Correction Factor | 0.1446 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | g/L | 0.7203 | 0.5488 | 13.2394 | 12.7283 | 0.0907 |
| CO2 | g/L | 71.3950 | 79.2075 | 102.0851 | 93.3947 | 58.1523 |
| HC (propane) | g/L | 0.2141 | 0.3557 | 1.8575 | 2.7752 | 0.2848 |
| appx PM mg | $\mathrm{mg} / \mathrm{L}$ | 6.7223 | 2.1627 | 38.6633 | 96.6982 | 0.2511 |


| Correction Factor | 0.2274 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO | $\mathrm{g} / \mathrm{L}$ | 1.0222 | 0.8232 | 9.5875 | 12.8843 | 0.0046 |
| CO2 | g/L | 152.4690 | 118.8113 | 82.6742 | 118.2473 | 85.7806 |
| HC (propane) | g/L | 0.8266 | 0.5335 | 1.7700 | 2.4806 | 1.1048 |
| appx PM mg | $\mathrm{mg} / \mathrm{L}$ | 5.7257 | 3.2441 | 7.5531 | 17.7353 | 0.6331 |

HOOD Results
Alcohol-

| Clean | Kero- |
| :--- | :--- |
| Cook | sene |


|  |  | Cook | sene | Solar |
| :---: | :---: | :---: | :---: | :---: |
| Totals |  | (cold start) |  |  |
| CO | grams | 2.71 | 5.43 |  |
| CO2 | grams | 306 | 409 |  |
| HC (propane) | grams (prop) | 0.6992 | 0.7374 |  |
| appx PM | mg | 2 | 2 |  |
| CO/CO2 ratio |  | 0.0138 | 0.0208 |  |
| Flame temp | degrees C | 6 | 479 |  |
|  |  |  |  |  |
| Totals |  | (hot start) |  |  |
| CO | grams | 2.71 | 5.06 |  |
| CO 2 | grams | 306 | 434 |  |
| HC (propane) | grams (prop) | 0.6992 | 1.6961 |  |
| appx PM gr | mg | 2 | 3 |  |
| CO/CO2 ratio |  | 0.0138 | 0.0185 |  |
| Flame temp | degrees C | 6 | 341 |  |
|  |  |  |  |  |
| Totals |  | (simmer) |  |  |
| CO | grams | 2.35 | 3.11 |  |
| CO 2 | grams | 350 | 403 |  |
| HC (propane) | grams (prop) | 1.3251 | 2.0206 |  |
| appx PM gr | grams | 2 | 5 |  |
| CO/CO2 ratio |  | 0.0106 | 0.0119 |  |
| Flame temp |  | 6 | 317 |  |

(Corrected for water temp and moisture)

| Correction Factor | 0.1408 |  |  |
| :---: | :---: | :---: | :---: |
| CO | g/L | 0.4722 | 0.8918 |
| CO2 | g/L | 53.0953 | 68.2801 |
| HC (propane) | g/L | 0.1195 | 0.1157 |
| appx PM mg | mg/L | 0.3072 | 0.4546 |


| Correction Factor | 0.1446 |  |  |
| :---: | :---: | :---: | :---: |
| CO | g/L | 0.4722 | 0.8347 |
| CO 2 | g/L | 53.0953 | 71.1966 |
| HC (propane) | g/L | 0.1195 | 0.2874 |
| appx PM mg | mg/L | 0.3072 | 0.5829 |


| Correction Factor | 0.2274 |  |  |
| :---: | :---: | :---: | :---: |
| CO | g/L | 0.5850 | 0.7004 |
| CO2 | g/L | 87.0337 | 94.5210 |
| HC (propane) | g/L | 0.3294 | 0.4944 |
| appx PM mg | mg/L | 0.5663 | 1.3829 |

HOOD Results

|  |  | 3 stone fire | Ghana Wood | 20 L Can Rocket | Mud/ Sawdust | VITA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Time to Boil (temp-corrected) | min | 26.69 | 21.84 | 22.29 | 15.99 | 14.00 |
| Temp-Corrected Specific Consumption | $\mathrm{g} / \mathrm{L}$ | 120.18 | 84.33 | 72.12 | 77.30 | 70.37 |
| Temp-Corr Specific Energy Consumption | kJ /L | 2,091.87 | 1,458.46 | 1,299.32 | 1,257.74 | 1,135.57 |
| Firepower | W | 8,001.96 | 6,490.30 | 5,670.31 | 7,902.64 | 8,036.66 |
| Thermal Efficiency | \% | 0.20 | 0.25 | 0.34 | 0.29 | 0.30 |

COOKING TASKS

| COTO Boil | g/L | 3.83 | 3.07 | 0.90 | 2.31 | 3.51 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO To Simmer | g/L | 7.30 | 7.01 | 2.17 | 7.39 | 5.05 |
| CO TO Cook | g/L | 11.14 | 10.08 | 3.06 | 9.71 | 8.56 |
|  |  | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| PM To Boil | mg/L | 257.64 | 533.95 | 128.91 | 295.00 | 337.81 |
| PM to Simmer | mg/L | 215.00 | 323.48 | 128.91 | 175.45 | 92.23 |
| PM to Cook | mg/L | 472.64 | 857.43 | 257.82 | 470.45 | 430.05 |
| CO2 to Boil | g/L | 121.12 | 88.45 | 122.98 | 75.79 | 71.88 |
| CO2 to S immer | g/L | 185.19 | 185.23 | 180.51 | 178.72 | 154.53 |
| CO2 to Cook | g/L | 306.31 | 273.68 | 303.49 | 254.51 | 226.41 |
| HC to Boil | g/L | 0.28 | 0.26 | 0.10 | 0.21 | 0.24 |
| HC to Simmer | g/L | 0.85 | 1.08 | 0.57 | 0.99 | 0.81 |
| HC to Cook | g/L | 1.13 | 1.33 | 0.67 | 1.19 | 1.05 |
|  |  |  |  |  |  |  |
| Average CO/CO2 Ratio for Boil |  | 0.0552 | 0.0564 | 0.0119 | 0.0501 | 0.0778 |
| CO/CO2 Ratio for Simmer |  | 0.0694 | 0.0594 | 0.0188 | 0.0645 | 0.0525 |
| Boiling |  |  |  |  |  |  |
| EF CO | g/kg | 51.40 | 70.37 | 14.48 | 42.78 | 93.47 |
| EF CO2 | g/kg | 1,623.05 | 2,031.03 | 2,000.21 | 1,429.53 | 1,910.97 |
| EF PM | $\mathrm{mg} / \mathrm{kg}$ | 3,519.59 | 12,884.23 | 1,976.11 | 5,123.86 | 8,327.68 |
| EF HC | g/kg | 3.79 | 6.00 | 1.61 | 3.85 | 6.41 |

HOOD Results
Patsari
Uganda 2- Proto-
Justa pot type Onil Eco-stove

| Time to Boil (temp-corrected) | min | 46.73 | 16.17 | 34.82 | 28.00 | 38.63 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Temp-Corrected Specific Consumption | $\mathrm{g} / \mathrm{L}$ | 132.43 | 52.35 | 111.54 | 116.87 | 234.23 |
| Temp-Corr Specific Energy Consumption | kJ /L | 2,221.74 | 800.88 | 1,666.19 | 1,708.19 | 4,490.09 |
| Firepower | W | 8,444.21 | 7,078.41 | 8,325.55 | 10,663.36 | 9,312.02 |
| Thermal Efficiency | \% | 0.19 | 0.43 | 0.22 | 0.20 | 0.15 |

COOKING TASKS
CO TO Boil
CO To Simmer
CO TO Cook
PM To Boil
PM to Simmer
PM to Cook
CO2 to Boil
CO2 to Simmer
CO2 to Cook
HC to Boil
HC to Simmer
HC to Cook

Average CO/CO2 Ratio for Boil CO/CO2 Ratio for Simmer Boiling
EF CO
EF CO2
EF PM EF HC
g/L
g/L
g/L
mg/L
mg/L
mg/L
g/L
g/L
g/L
g/L
g/L
g/L
g/kg
g/kg
$\mathrm{mg} / \mathrm{kg}$ g/kg

| 1.96 | 1.17 | 1.18 | 2.04 | 6.06 |
| ---: | ---: | ---: | ---: | ---: |
| 2.87 | 3.28 | 2.70 | 4.27 | 3.55 |
| 4.83 | 4.46 | 3.89 | 6.30 | 9.61 |
| 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| 80.87 | 67.43 | 70.59 | 111.89 | 747.05 |
| 77.61 | 68.20 | 105.31 | 156.62 | 273.34 |
| 158.48 | 135.63 | 175.90 | 268.51 | $1,020.39$ |
|  |  |  |  |  |
| 163.08 | 34.01 | 129.99 | 124.38 | 288.54 |
| 280.58 | 166.64 | 224.63 | 269.02 | 266.74 |
| 443.67 | 200.64 | 354.62 | 393.40 | 555.28 |
|  |  |  |  |  |
| 0.45 | 0.16 | 0.24 | 0.22 | 0.55 |
| 1.31 | 1.24 | 0.93 | 1.00 | 1.07 |
| 1.76 | 1.40 | 1.16 | 1.22 | 1.62 |
|  |  |  |  |  |
|  |  |  |  |  |
| 0.0174 | 0.0534 | 0.0147 | 0.0239 | 0.0325 |
| 0.0168 | 0.0304 | 0.0205 | 0.0247 | 0.0201 |
|  |  |  |  |  |
| 27.60 | 51.47 | 24.06 | 37.70 | 43.21 |
| $2,348.28$ | $1,497.56$ | $2,660.75$ | $2,397.11$ | $2,088.72$ |
| $1,219.54$ | $3,005.25$ | $1,425.69$ | $2,310.11$ | $5,527.62$ |
| 6.60 | 7.16 | 4.89 | 4.47 | 4.05 |

HOOD Results

|  |  | Wood Flame Fan | Wood Gas Fan | Mali Char coal | Charcoal | Propane |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Time to Boil (temp-corrected) | min | 19.50 | 23.75 | 38.62 | 28.35 | 22.98 |
| Temp-Corrected Specific Consumption | g/L | 49.76 | 46.91 | 81.14 | 68.30 | 12.82 |
| Temp-Corr Specific Energy Consumption | kJ/L | 836.17 | 754.73 | 2,200.86 | 1,928.27 | 590.92 |
| Firepower | W | 4,047.99 | 2,708.92 | 5,650.63 | 6,262.64 | 1,930.36 |
| Thermal Efficiency | \% | 0.42 | 0.45 | 0.18 | 0.18 | 0.68 |


| CO TO Boil | g/L | 0.82 | 0.55 | 12.97 | 14.15 | 0.10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CO To Simmer | g/L | 1.02 | 0.82 | 9.59 | 12.88 | 0.00 |
| CO TO Cook | g/L | 1.84 | 1.37 | 22.56 | 27.04 | 0.11 |
|  |  | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| PM To Boil | mg/L | 3.79 | 2.16 | 44.46 | 99.62 | 0.27 |
| PM to Simmer | $\mathrm{mg} / \mathrm{L}$ | 5.73 | 3.24 | 7.55 | 17.74 | 0.63 |
| PM to Cook | $\mathrm{mg} / \mathrm{L}$ | 9.52 | 5.41 | 52.01 | 117.35 | 0.90 |
| CO 2 to Boil | g/L | 71.65 | 79.21 | 96.08 | 95.99 | 54.23 |
| CO2 to Simmer | g/L | 152.47 | 118.81 | 82.67 | 118.25 | 85.78 |
| CO2 to Cook | g/L | 224.12 | 198.02 | 178.76 | 214.24 | 140.01 |
| HC to Boil | g/L | 0.19 | 0.36 | 1.61 | 1.96 | 0.23 |
| HC to Simmer | g/L | 0.83 | 0.53 | 1.77 | 2.48 | 1.10 |
| HC to Cook | g/L | 1.02 | 0.89 | 3.38 | 4.44 | 1.33 |
|  |  |  |  |  |  |  |
| Average CO/CO2 R atio for Boil |  | 0.0192 | 0.0109 | 0.2249 | 0.2050 | 0.0034 |
| CO/CO2 Ratio for Simmer |  | 0.0117 | 0.0109 | 0.2251 | 0.1726 | 0.0001 |
| Boiling |  |  |  |  |  |  |
| EF CO | g/kg | 21.42 | 17.59 | 284.53 | 390.40 | 8.62 |
| EF CO2 | g/kg | 1,863.01 | 2,538.40 | 2,091.55 | 2,658.17 | 4,542.56 |
| EF PM | $\mathrm{mg} / \mathrm{kg}$ | 98.36 | 69.31 | 4,131.65 | 5,834.94 | 21.85 |
| EF HC | g/kg | 5.07 | 11.40 | 35.82 | 55.63 | 18.95 |

HOOD Results

|  |  | Clean Cook | Kerosene | Solar |
| :---: | :---: | :---: | :---: | :---: |
| Time to Boil (temp-corrected) | min | 31.62 | 41.89 | 69.95 |
| Temp-Corrected Specific Consumption | g/L | 28.68 | 23.00 |  |
| Temp-Corr S pecific Energy Consumption | kJ/L | 612.83 | 913.49 |  |
| Firepower | W | 1,543.64 | 1,917.90 | 2,236.41 |
| Thermal Efficiency | \% | 0.66 | 0.52 | 0.25 |


| COTO Boil | g/L | 0.47 | 0.86 | 0.00 |
| :---: | :---: | :---: | :---: | :---: |
| CO To Simmer | g/L | 0.59 | 0.70 | 0.00 |
| CO TO Cook | g/L | 1.06 | 1.56 | 0.00 |
|  |  |  | 0.00 | 0.00 |
| PM To Boil | mg/L | 0.31 | 0.52 | 0.00 |
| PM to Simmer | mg/L | 0.57 | 1.38 | 0.00 |
| PM to Cook | mg/L | 0.87 | 1.90 | 0.00 |
| CO2 to Boil | g/L | 53.10 | 69.74 | 0.00 |
| CO2 to Simmer | g/L | 87.03 | 94.52 | 0.00 |
| CO2 to Cook | g/L | 140.13 | 164.26 | 0.00 |
|  |  |  |  | 0.00 |
| HC to Boil | g/L | 0.12 | 0.20 | 0.00 |
| HC to Simmer | g/L | 0.33 | 0.49 | 0.00 |
| HC to Cook | g/L | 0.45 | 0.70 | 0.00 |
|  |  |  |  |  |
|  |  |  |  |  |
| Average CO/CO2 Ratio for Boil |  | 0.0138 | 0.0197 |  |
| CO/CO2 R atio for Simmer |  | 0.0106 | 0.0119 |  |
| Boiling |  |  |  |  |
| EF CO | g/kg | 16.41 | 46.24 |  |
| EF CO2 | g/kg | 1,852.21 | 3,714.77 |  |
| EF PM | $\mathrm{mg} / \mathrm{kg}$ | 10.87 | 24.81 |  |
| EF HC | g/kg | 4.24 | 10.74 |  |

## WBT Results

## AVERAGES

## Chimney

|  | 3 stone <br> fire | Ghana <br> Wood | 20L Can <br> Rocket | Mud I <br> Sawdust | VITA |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | Justa

## Standard Deviations

Chimney

|  | 3 stone <br> fire | Ghana <br> Wood | 20L Can <br> Rocket | Mud/Saw <br> dust | VITA | Justa |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |

## WBT Results

| AVERAGES | Chimney Stoves |  |  |  | Electric Fan |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Uganda 2pot | Patsari <br> Proto- <br> type | Onil | Ecostove | Wood Flame Fan | Wood Gas Fan |
| 1. HIGH POWER TEST (COLD START) | Average | Average | Average | Average | Average | Average |
| Time to boil Pot \# 1 | 15.00 | 30.93 | 33.17 | 52.23 | 22.33 | 31.50 |
| Burning rate | 25.51 | 31.15 | 38.28 | 29.59 | 13.96 | 8.24 |
| Thermal efficiency | 0.36 | 0.20 | 0.16 | 0.21 | 0.40 | 0.44 |
| Specific fuel consumption | 64.62 | 118.89 | 149.06 | 193.44 | 63.83 | 53.68 |
| Temp-corrected specific consumption | 60.15 | 110.16 | 133.10 | 167.23 | 52.01 | 48.74 |
| Firepower | 8,288 | 10,018 | 12,439 | 7,733 | 4,535 | 2,678 |
| 2. HIGH POWER TEST (HOT START) | Average | Average | Average | Average | Average | Average |
| Time to boil Pot \# 1 | 12.83 | 29.33 | 22.53 | 36.77 | 23.50 | 28.90 |
| Burning rate | 27.06 | 31.36 | 41.09 | 29.96 | 12.51 | 8.50 |
| Thermal efficiency | 0.39 | 0.21 | 0.19 | 0.24 | 0.41 | 0.46 |
| Specific fuel consumption | 71.52 | 191.81 | 192.76 | 235.03 | 60.68 | 51.05 |
| Temp-corrected specific consumption | 69.46 | 176.12 | 174.32 | 206.79 | 49.83 | 46.60 |
| Firepower | 8,794 | 10,087 | 13,354 | 7,830 | 4,065 | 2,761 |
| 3. LOW POWER (SIMMER) | Average | Average | Average | Average | Average | Average |
| Burning rate | 8.79 | 12.14 | 16.24 | 14.08 | 6.84 | 4.03 |
| Thermal efficiency | 0.31 | 0.14 | 0.12 | 0.19 | 0.39 | 0.46 |
| Specific fuel consumption | 105.59 | 133.44 | 182.16 | 162.54 | 81.66 | 44.85 |
| Firepower | 3,050 | 4,170 | 5,635 | 3,988 | 2,372 | 1,400 |
| Turn down ratio | 2.93 | 2.46 | 2.37 | 2.03 | 1.72 | 1.97 |

## Standard Deviations

Chimney Stoves

|  | Electric Fan |
| :--- | :--- |
|  | Wood |
| Flame | Wood |
| Ecostove | Fan |$\quad$ GasFan


| 1. HIGH POWER TEST (COLD START) | St Dev | St Dev |  | St Dev |  | St Dev |  |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: |


| 2. HIGH POWER TEST (HOT START) | St Dev | St Dev | St Dev | St Dev | St Dev | StDev |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Time to boil Pot \# 1 | 1.89 | 4.35 | 0.46 | 2.12 | 1.80 | 2.62 |
| Burning rate | 4.32 | 3.03 | 2.39 | 2.34 | 1.23 | 0.37 |
| Thermal efficiency | 0.03 | 0.01 | 0.01 | 0.02 | 0.01 | 0.01 |
| Specific fuel consumption | 3.97 | 13.87 | 6.76 | 6.87 | 1.90 | 2.79 |
| Temp-corrected specific consumption | 5.74 | 16.42 | 6.36 | 14.08 | 2.47 | 2.62 |
| Firepower | 1,403 | 979 | 775 | 611 | 400 | 121 |


| 3. LOW POWER (SIMMER) | St Dev | St Dev | St Dev | St Dev | St Dev | StDev |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Burning rate | 0.86 | 1.45 | 0.47 | 2.84 | 0.36 | 0.12 |
| Thermal efficiency | 0.02 | 0.01 | 0.02 | 0.04 | 0.03 | 0.02 |
| Specific fuel consumption | 12.24 | 15.97 | 2.16 | 32.33 | 3.83 | 1.73 |
| Firepower | 297 | 502 | 162 | 804 | 127 | 43 |
| Turn down ratio | 0.74 | 0.50 | 0.07 | 0.53 | 0.23 | 0.10 |

## WBT Results

| AVERAGES | Charcoal |  | Other Fuels |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | Mali <br> Charcoal | Gyapa Charcoal | Propane | Alcohol - <br> Clean <br> Cook* | Kerosene |
| 1. HIGH POWER TEST (COLD START) | Average | Average | Average | Average | Average |
| Time to boil Pot \# 1 | 36.70 | 29.77 | 30.58 |  | 51.57 |
| Burning rate | 13.76 | 13.58 | 2.45 |  | 2.65 |
| Thermal efficiency | 0.15 | 0.17 | 0.66 |  | 0.59 |
| Specific fuel consumption | 107.43 | 83.75 | 14.57 |  | 31.90 |
| Temp-corrected specific consumption | 96.31 | 76.32 | 12.61 |  | 28.05 |
| Firepower | 7,267 | 7,169 | 1,986 |  | 1,920 |
| 2. HIGH POWER TEST (HOT START) | Average | Average | Average | Average | Average |
| Time to boil P ot \# 1 | 42.93 | 33.40 | 32.00 |  | 50.67 |
| Burning rate | 11.49 | 11.97 | 2.38 |  | 3.05 |
| Thermal efficiency | 0.16 | 0.18 | 0.62 |  | 0.52 |
| Specific fuel consumption | 104.58 | 82.83 | 15.89 |  | 36.77 |
| Temp-corrected specific consumption | 96.17 | 76.21 | 13.84 |  | 32.16 |
| Firepower | 6,066 | 6,323 | 1,934 |  | 2,209 |
| 3. LOW POWER (SIMMER) | Average | Average | Average | Average | Average |
| Burning rate | 4.00 | 8.50 | 1.23 |  | 2.91 |
| Thermal efficiency | 0.28 | 0.18 | 0.61 |  | 0.37 |
| Specific fuel consumption | 46.04 | 102.49 | 13.75 |  | 35.74 |
| Firepower | 2,200 | 4,677 | 1,028 |  | 2,256 |
| Turn down ratio | 2.78 | 1.53 | 1.93 |  | 0.98 |

* Initial WBT tests of the Alcohol- Clean Cook stove were dismissed after receiving an improved model in Dec. 2006


## Standard Deviations

Charcoal
Other Fuels

## Alcohol -

|  | Mali <br> Charcoal | Gyapa <br> Charcoal |  | Clean <br> Propane <br> Cook* $^{\prime}$ | Kerosene |
| :--- | ---: | ---: | ---: | ---: | ---: |
| 1. HIGH POWER TEST (COLD START) | St Dev | St Dev | St Dev | St Dev | St Dev |
| Time to boil Pot \# 1 | 2.29 | 2.36 | 9.36 | 4.99 |  |
| Burning rate | 0.23 | 1.13 | 0.69 | 0.12 |  |
| Thermal efficiency | 0.01 | 0.01 | 0.02 | 0.22 |  |
| Specific fuel consumption | 6.27 | 7.29 | 0.90 | 1.79 |  |
| Temp-corrected specific consumption | 5.35 | 7.02 | 0.43 | 1.70 |  |
| Firepower | 120 | 599 | 561 | 84 |  |
| 2. HIGH POWER TEST (HOT START) | St Dev | St Dev | St Dev | St Dev | St Dev |
| Time to boil Pot \# 1 | 5.29 | 3.86 | 3.00 | 3.62 |  |
| Burning rate | 0.94 | 1.52 | 0.13 | 0.65 |  |
| Thermal efficiency | 0.01 | 0.02 | 0.06 | 0.11 |  |
| Specific fuel consumption | 3.82 | 4.61 | 0.75 | 11.58 |  |
| Temp-corrected specific consumption | 5.31 | 7.81 | 1.15 | 10.76 |  |
| Firepower | 494 | 803 | 107 | 471 |  |
| 3. LOW POWER (SIMMER) | St Dev | St Dev | St Dev | St Dev | St Dev |
| Burning rate | 0.50 | 2.96 | 0.25 | 0.15 |  |
| Thermal efficiency | 0.04 | 0.03 | 0.10 | 0.05 |  |
| Specific fuel consumption | 5.99 | 40.39 | 2.84 | 1.83 |  |
| Firepower | 275 | 1627 | 208 | 113 |  |
| Turn down ratio | 0.31 | 0.80 | 0.36 | 0.19 |  |

*Initial WBT tests of the Alcohol- Clean Cook stove were dismissed after receiving an improved model in Dec. 2006

## TEST KITCHEN Results

## AVERAGES

|  | 3 stone <br> fire | Ghana <br> Wood | 20 L Can <br> Rocket | Mud/Saw- <br> dust | VITA |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
|  |  |  |  |  |  |
| 1. HIGH POWER TEST (COLD START) | Average | Average | Average | Average | Average |
| Time to boil Pot \# 1 | 36.00 | 20.99 | 16.27 | 15.97 | 18.33 |
| Burning rate | 17.08 | 26.58 | 21.33 | 33.10 | 31.82 |
| Thermal efficiency | 0.22 | 0.21 | 0.31 | 0.22 | 0.20 |
| Specific fuel consumption | 125.76 | 114.76 | 70.57 | 107.80 | 118.80 |
| Temp-corrected specific consumption | 103.04 | 95.98 | 64.85 | 94.82 | 98.73 |
| Firepower | 5,549 | 8,637 | 6,930 | 10,755 | 10,339 |
| 3. LOW POWER (SIMMER) | Average | Average | Average | Average | Average |
| Burning rate | 10.81 | 7.88 | 6.59 | 4.73 | 11.25 |
| Thermal efficiency | 0.27 | 0.37 | 0.25 | 0.57 | 0.45 |
| Specific fuel consumption | 82.22 | 58.99 | 45.15 | 34.03 | 96.28 |
| Firepower | 3,752 | 2,734 | 2,288 | 1,640 | 3,902 |
| Turn down ratio | - | - | - | - | - |

## STANDARD Deviations

|  | 3 stone <br> fire | Ghana <br> Wood | 20 L Can <br> Rocket | Mud/Saw- <br> dust | VITA |
| :--- | :---: | :---: | :---: | :---: | :---: | ---: | ---: |

## TEST KITCHEN Results

| AVERAGES | Electric Fan |  | Other Fuels |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | Wood <br> Flame <br> Fan | Wood Gas Fan | Propane | Alcohol - <br> Clean <br> Cook | Kerosene |
| 1. HIGH POWER TEST (COLD START) | Average | Average | Average | Average | Average |
| Time to boil Pot \# 1 | 22.00 | 27.94 | 47.50 | 59.33 | 35.38 |
| Burning rate | 14.65 | 9.17 | 1.56 | 3.67 | 3.33 |
| Thermal efficiency | 0.37 | 0.45 | 0.76 | 0.55 | 0.47 |
| Specific fuel consumption | 67.64 | 53.61 | 16.60 | 49.66 | 22.99 |
| Temp-corrected specific consumption | 55.30 | 49.04 | 13.38 | 41.43 | 20.24 |
| Firepower | 4,759 | 2,979 | 1,265 | 1,218 | 2,415 |
| 3. LOW POWER (SIMMER) | Average | Average | Average | Average | Average |
| Burning rate | 5.73 | 4.37 | 1.30 | 4.00 | 2.71 |
| Thermal efficiency | 0.50 | 0.45 | 0.63 | 0.53 | 0.27 |
| Specific fuel consumption | 42.47 | 30.98 | 9.36 | 29.06 | 17.27 |
| Firepower | 1,988 | 1,518 | 1,084 | 1,513 | 2,097 |
| Turn down ratio | - | - | - | - | - |
| STANDARD Deviations | Electric Fan |  | Other Fuels |  |  |
|  | Wood <br> Flame <br> Fan | Wood Gas Fan | Propane | Alcohol - <br> Clean <br> Cook | Kerosene |
| 1. HIGH POWER TEST (COLD START) | St Dev | St Dev | St Dev | St Dev | St Dev |
| Time to boil Pot \# 1 | 2.83 | 0.92 | 9.19 | 10.07 | 12.90 |
| Burning rate | 0.01 | 0.44 | 0.41 | 0.27 | 1.22 |
| Thermal efficiency | 0.04 | 0.05 | 0.30 | 0.00 | 0.01 |
| Specific fuel consumption | 8.90 | 1.92 | 7.67 | 7.79 | 0.10 |
| Temp-corrected specific consumption | 7.03 | 3.07 | 6.37 | 3.67 | 0.11 |
| Firepower | 2 | 143 | 335 | 88 | 881 |
| 3. LOW POWER (SIMMER) | St Dev | St Dev | St Dev | St Dev | St Dev |
| Burning rate | 0.21 | 0.60 | 0.05 | - | 0.06 |
| Thermal efficiency | 0.00 | 0.02 | 0.02 | 0.04 | 0.12 |
| Specific fuel consumption | 1.45 | 4.55 | 0.52 | 0.10 | 2.38 |
| Firepower | 74 | 209 | 39 | 0 | 43 |
| Turn down ratio | - | - | - | - | - |

## Overall Variations

|  |  |  | \# of <br> Tests | 3 stone <br> fire | Ghana <br> Wood | 20 L Can <br> Rocket | Mud/ <br> Sawdust | VITA |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

C oefficient of Variation = Standard Deviation / Average
Includes different testers, different times of year.
Most of the CO variability is in the chimney stoves
Due to problems in test phases of the Wood Gas and Alcohol Stoves, variation are not reported

|  | Justa | Uganda <br> 2-pot | Patsari <br> proto- <br> type | Onil | Eco- <br> stove | Wood <br> Flame <br> Fan | Wood <br> Gas <br> Fan |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Mali <br> Char- <br> coal |  |  |  |  |  |
| Cold Temp Corr Time to Boil | $5 \%$ | $30 \%$ | $2 \%$ | $30 \%$ | $17 \%$ | $9 \%$ | $18 \%$ |
| Hot Temp Corr Time to Boil | $26 \%$ | $12 \%$ | $25 \%$ | $35 \%$ | $13 \%$ | $14 \%$ | $54 \%$ |
|  |  |  |  |  |  |  |  |
| Cold Firepower | $33 \%$ | $24 \%$ | $26 \%$ | $16 \%$ | $14 \%$ | $14 \%$ | $23 \%$ |
| Hot Firepower | $22 \%$ | $24 \%$ | $25 \%$ | $27 \%$ | $29 \%$ | $8 \%$ | $32 \%$ |
| Simmer Firepower | $5 \%$ | $19 \%$ | $22 \%$ | $16 \%$ | $28 \%$ | $12 \%$ | $44 \%$ |
|  |  |  |  |  |  |  |  |
| Cold Specific Consumption | $14 \%$ | $16 \%$ | $16 \%$ | $11 \%$ | $32 \%$ | $14 \%$ | $20 \%$ |
| Hot Specific Consumption | $27 \%$ | $30 \%$ | $46 \%$ | $45 \%$ | $30 \%$ | $4 \%$ | $16 \%$ |
| Simmer Specific Consumption | $6 \%$ | $16 \%$ | $20 \%$ | $13 \%$ | $24 \%$ | $13 \%$ | $43 \%$ |
| Cold CO per L |  |  |  |  |  |  |  |
| Hot CO per L | $73 \%$ | $49 \%$ | $52 \%$ | $65 \%$ | $2 \%$ | $26 \%$ | $22 \%$ |
| Simmer CO per L | $79 \%$ | $23 \%$ | $36 \%$ | $56 \%$ | $3 \%$ | $87 \%$ | $13 \%$ |

Coefficient of Variation = Standard Deviation / Average
Includes different testers, different times of year.
Most of the CO variability is in the chimney stoves
Due to problems in test phases of the Wood Gas and Alcohol Stoves, variation are not reported

## Overall Variations

|  | Gyapa <br> Char- <br> coal | Pro- <br> pane | Alcohol- <br> Clean <br> Cook |
| :--- | :---: | :---: | ---: |
| Kero- <br> sene |  |  |  |
|  |  |  |  |
| Cold Temp Corrected Time to Boil | $12 \%$ | $3 \%$ | $12 \%$ |
| Hot Temp Corrected Time to Boil | $12 \%$ | $10 \%$ | $1 \%$ |
|  |  |  | $14 \%$ |
| Cold Firepower | $21 \%$ | $24 \%$ | $14 \%$ |
| Hot Firepower | $12 \%$ | $5 \%$ | $21 \%$ |
| Simmer Firepower | $51 \%$ | $8 \%$ | $17 \%$ |
|  |  |  | $24 \%$ |
| Cold Specific Consumption | $7 \%$ | $4 \%$ | $24 \%$ |
| Hot Specific Consumption | $14 \%$ | $6 \%$ |  |
| Simmer Specific Consumption | $50 \%$ | $11 \%$ | $10 \%$ |
|  |  |  | $0 \%$ |
| Cold CO per L | $17 \%$ | $41 \%$ | $62 \%$ |
| Hot CO per L | $8 \%$ | $5 \%$ |  |
| Simmer CO per L | $7 \%$ | $22 \%$ |  |

## Solar Cooker Tests



| Test 1 | 8/22/2004 |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | start | end | Firepower |  |  |
| Cold start | 12:30 | 1:27 | Minutes |  |  |
|  | 17.4 | 99.2 | Temp |  |  |
|  | 5,840 | 5,610 | g water |  |  |
|  |  |  | Firepower | 2,656 W hot start | 57 min |
|  |  |  | 28\% Efficiency cold start |  |  |
| Hot start | 1:47 | 3:10 | Minutes |  |  |
|  | 18.1 | 99.2 | Temp |  |  |
|  | 5,840.0 | 5,582.0 | g water |  |  |
|  |  |  | Firepower | 2,528 W cold star | 83 min |
|  |  |  | 20\% Efficiency hot start |  |  |
| Simmer | 3:10 | 3:55 | Minutes |  |  |
|  |  |  | Firepower | 1,383 W simmer | 45 min |



## Test 3 8/24/2004

|  | start | end |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Cold start | 12:12 | 1:52 | Minutes |  |  |
|  | 16.1 | 99.2 | Temp |  |  |
|  | 5,840 | 5,614 | g water |  |  |
|  |  |  | Firepower | 1,509 W hot start | 100 min |
|  |  |  |  | 28\% Efficiency c |  |
| Hot start | 1:52 | 3:07 | Minutes |  |  |
|  | 15.4 | 99.2 | Temp |  |  |
|  | 5,840 | 5,690 | g water |  |  |
|  |  |  | Firepower | 2,798 W cold star | 75 min |
|  |  |  |  | 19\% Efficiency h |  |
| Simmer | 3:07 | 3:52 | Minutes |  |  |
|  |  |  | Firepower | 1,383 W simmer | 45 min |


| Average cold start firepower | $\mathbf{2 , 0 8 7} \mathbf{~ W}$ | $\mathbf{7 6}$ min |
| :--- | :--- | :--- |
| Average hot start firepower | $\mathbf{2 , 3 8 6} \mathbf{~ W}$ | $\mathbf{7 7}$ min |
| Average simmer firepower | $1,383 \mathrm{~W}$ | $\mathbf{4 5}$ min |
|  |  |  |
| Average cold start Efficiency | $21 \%$ |  |
| Average hot start efficiency | $23 \%$ |  |


| Average boil firepower | 2,236 Watts |
| :---: | :---: |
| Average boil efficiency | $22 \%$ |

## Safety Ratings

A method for evaluating safety, proposed by Nathan Johnson ${ }^{10}$ of Iowa State University, was used to evaluate safety in these stoves, in this case without the weighted rankings his system suggests. Each of the following criteria were rated as excellent (4 points), good (3 points), fair ( 2 points) or poor (1 point) for safety evaluation.

## Results of Evaluation:

## No. Name

1 Sharp Edges/Points
2 Cookstove Tipping
3 Containment of Combustion
4 Expulsion of Fuel
5 Obstructions Near Cooking Surface
6 Surface Temperature
7 Heat Transfer to Surroundings
8 Cookstove Handle Temperature
9 Flames/Heat Surrounding Cookpot
10 Flames/Head Exiting Fuel Chanber

| Stove | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ | $\mathbf{6}$ | $\mathbf{7}$ | $\mathbf{8}$ | $\mathbf{9}$ | $\mathbf{1 0}$ | Total |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Onil | 4 | 4 | 4 | 4 | 4 | 3 | 4 | 4 | 4 | 4 | 39 |
| Patsari Prototype | 2 | 4 | 4 | 4 | 2 | 4 | 4 | 4 | 4 | 4 | 36 |
| Justa | 4 | 4 | 4 | 4 | 2 | 4 | 4 | 4 | 4 | 4 | 38 |
| Ecostove | 2 | 3 | 4 | 4 | 3 | 2 | 4 | 4 | 4 | 4 | 34 |
| Uganda 2-pot | 4 | 3 | 4 | 4 | 4 | 2 | 4 | 4 | 4 | 4 | 37 |
| Wood Flame | 4 | 2 | 3 | 4 | 4 | 3 | 4 | 4 | 3 | 4 | 35 |
| Propane | 4 | 1 | 4 | 4 | 4 | 4 | 4 | 4 | 2 | 2 | 33 |
| Kerosene | 4 | 2 | 3 | 4 | 3 | 3 | 4 | 4 | 4 | 4 | 35 |
| Alcohol | 4 | 3 | 4 | 4 | 4 | 4 | 4 | 4 | 3 | 3 | 37 |
| Mali Charcoal | 4 | 2 | 3 | 4 | 3 | 2 | 4 | 3 | 4 | 4 | 33 |
| Wood Gas | 4 | 1 | 3 | 4 | 4 | 3 | 4 | 2 | 4 | 4 | 33 |
| Mud/Sawdust | 4 | 2 | 2 | 3 | 4 | 3 | 4 | 4 | 3 | 4 | 33 |
| 20L Can Rocket | 2 | 2 | 3 | 4 | 4 | 2 | 4 | 4 | 4 | 4 | 33 |
| Ghana Wood | 4 | 2 | 2 | 3 | 4 | 2 | 4 | 4 | 3 | 4 | 32 |
| Ghana Charcoal | 4 | 2 | 2 | 3 | 4 | 2 | 4 | 3 | 4 | 4 | 32 |
| VITA | 2 | 2 | 3 | 3 | 4 | 1 | 3 | 4 | 3 | 4 | 29 |
| 3 Stone fire | 4 | 1 | 2 | 1 | 4 | 1 | 2 | 4 | 1 | 1 | 21 |
| Solar Cooker* | 2 | 4 | 4 | 4 | 4 | 2 | 1 | 4 | 4 | 3 | 32 |

${ }^{(*)}$ Even though no flames are present, focal point solar cooker is extremely hot when uncovered and spontaneous combustion may occur if not careful.

[^3]
## Safety Evaluation Procedures

For further details on this safety evaluation method go to http://www.vrac.iastate.edu/~atlas/safety.htm.

| Stove <br> Tester | Location |
| :--- | :--- |

## 1. SHARP EDGES AND POINTS

Equipment: Cloth, rag, or loose clothing

## Procedure:

a) Rub cloth along exterior surfaces.
b) Note number of times cloth catches / tears.

| Rating | No. of catches |  |
| :---: | :---: | :--- |
| Poor (1) | four or more | No. _-_ |
| Fair (2) | three |  |
| Good (3) | one or two | Result 1 |
| Best (4) | none |  |

## 2. COOKSTOVE TIPPING

Equipment: Fuel, ruler / tape measure, calculator

## Procedure:

a) Set stove on flat surface and load with fuel but do not ignite.
b) Pick a side to tip towards and measure the height of its tallest point, place value into Table $A$.
c) Slowly tip cookstove in the outward direction from the side chosen until the stove begins to tip on its own.
d) Hold stove tilted where it can overturn and measure new height of the point chosen in part 'b', place value into Table A.
e) Using a calculator, divide the tipped height by the standing height to find the ratio $R$, place into Table $A$.
f) Repeat process as many times as there are legs on the stove (or four times for a circular base).
g) Use the largest ratio in Table $A$ with the metric in Table B to find the most deficient rating for the result.

| Run | Starting <br> Height | Tipped <br> Height | Ratio |
| :---: | :---: | :---: | :---: |
| 1 | - | - | - |
| 2 | - | - | - |
| 3 | - | - | - |
| 4 | - | - | - |
| 5 | - | - | - |
| 6 | - | - | - |


| Rating | No. of catches |
| :---: | :---: |
| Poor (1) | $\mathrm{R}>0.978$ |
| Fair (2) | $0.961<\mathrm{R}<0.978$ |
| Good (3) | $0.940<\mathrm{R}<0.961$ |
| Best (4) | $\mathrm{R}<0.940$ |
| Result 2 |  |

## 3. CONTAINMENT OF FUEL

(solar stoves receive Best rating)
Equipment: Fuel, ruler / tape measure, cookpot

## Procedure:

a) The cookstove should be stocked with fuel but not ignited.
b) Place cookpot onto burner.
c) Sum approximate areas through which fuel can be seen.
d) Use the summation of area, $A$, to find the rating.

| Rating | No. of catches |
| :---: | :---: |
| Poor | $\mathrm{A} \geq 250$ |
| Fair | $150 \leq \mathrm{A}<250$ |
| Good | $50 \leq \mathrm{A}<150$ |
| Best | $\mathrm{A}<50$ |

Area
Notes:


## 4. OBSTRUCTIONS NEAR COOKING SURFACE

$($ skirt-stove $=$ Good; solar $=$ Best $)$
Equipment: Ruler / tape measure

## Procedure:

a) Inspect cookstove for skirt, do not perform if skirt is present.
b) Measure height difference between the cooking surface and obstructions surrounding the cooking surface.
c) Use the largest height difference, $D$, to find the rating.

Notes:

| Rating | No. of catches |
| :---: | :---: |
| Poor | $\mathrm{D} \geq 4$ |
| Fair | $2.5 \leq \mathrm{D}<4$ |
| Good | $1 \leq \mathrm{D}<2.5$ |
| Best | $\mathrm{D}<1$ |

Largest


## 5. SURFACE TEMPERATURE; 6. HEAT TRANSMISSION TO SURROUNDINGS;

Equipment: Fuel, igniter, chalk, ruler / tape measure, hand-held thermocouple

## Procedure:

a) Chalk $8 \times 8 \mathrm{~cm}$ grid onto cookstove and also within an outline of cook stove on the floor if within 5 cm of undercarriage, and within an outline of cookstove onto the wall if within 10 cm , while continuing the grid 16 cm higher up the wall than the top of the cookstove, if stove is mounted to floor or wall, take supplementary wall and floor temperatures by using cookstove surface temperature near where it attaches to floor or wall. b) Chalk extra thick lines at 0.9 m and 1.5 m onto cookstove, if applicable. c) Ignite fuel and continue up to step ' $g$ ' then wait at that step until cookstove has reached max temp ( -20 min ) before proceeding, adding fuel when necessary. d) Use the largest height difference, $D$, to find the rating. e) Measure air temp. f) Compute values for Tables $B$ by adding air temp to temps located in Tables $A$. g) Take data using thermocouple at grid intersections. h) Start with wall and floor by moving cookstove away to take measurements for up to one minute, then return cookstove for at least 5 minutes, taking surface temp and operational construction temp data while waiting, repeat step 'h' until all data points have been checked. i) Find maximum temperatures for all scenarios. j) Find which rating is given by the maximum temperature using Tables B. k) Use most deficient ratings for the results.


| HEAT TRANSFER TO THE ENVIRONMENT |  |  |  | HANDLE TEMPERATURE |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Rating | Floor | Wall |  | ting | Metallic | Nonmetallic |
| 6A | Poor Fair Good Best | $\begin{aligned} & \mathrm{T} \geq 65 \\ & 55 \leq \mathrm{T}<65 \\ & 45 \leq \mathrm{T}<55 \\ & \mathrm{~T}<45 \end{aligned}$ | $\begin{gathered} \mathrm{T} \geq 80 \\ 70 \leq \mathrm{T}<80 \\ 60 \leq \mathrm{T}<70 \\ \mathrm{~T}<60 \end{gathered}$ | 7A | Poor <br> Fair <br> Good <br> Best | $\begin{aligned} \mathrm{T} & \geq 32 \\ 26 & \leq \mathrm{T}<32 \\ 20 & \leq \mathrm{T}<26 \\ \mathrm{~T} & <20 \end{aligned}$ | $\begin{gathered} \mathrm{T} \geq 44 \\ 38 \leq \mathrm{T}<44 \\ 32 \leq \mathrm{T}<38 \\ \mathrm{~T}<32 \end{gathered}$ |
| 6B | Poor <br> Fair <br> Good <br> Best | $\begin{gathered} \mathrm{T} \geq- \\ -\leq \mathrm{T}<- \\ -\mathrm{T}<- \\ \mathrm{T}< \\ < \end{gathered}$ | $\begin{gathered} \mathrm{T} \geq- \\ \leq \mathrm{T}<- \\ -\mathrm{T}<- \\ \mathrm{T}<- \end{gathered}$ | 7B | Poor <br> Fair <br> Good <br> Best | $\begin{gathered} \mathrm{T} \geq \\ -\mathrm{T}< \\ -\leq \mathrm{T}<- \\ \mathrm{T}< \end{gathered}$ | $\begin{gathered} \mathrm{T} \geq- \\ -\leq \mathrm{T}<-- \\ -\leq \mathrm{T}<-\overline{2} \\ \mathrm{~T}<- \end{gathered}$ |
| Max/Rating |  | _ $/$ | - 1 |  | /Rating | _ | _ $/$ |


| Result 5 |  |
| :--- | :--- |



Notes:

## - - - - - - - - - - - - - - - - - -

## 8. CHIMNEY SHIELDING

(solar stoves and stoves without chimneys receive Best rating)
Equipment: Fuel, igniter, chalk, ruler / tape measure, hand-held thermocouple

## Procedure:

a) If the chimney has no protective shielding, surface temperature metrics from Test 5 are used for rating.
b) If the chimney has protective covering, measurements are taken to calculate the average area of gaps, $A$.

| Rating | Hole size $\left(\mathbf{c m}^{2}\right)$ |
| :---: | :---: |
| Poor | $\mathrm{A} \geq 150$ |
| Fair | $50 \leq \mathrm{A}<150$ |
| Good | $10 \leq \mathrm{A}<50$ |
| Best | $\mathrm{A}<10$ |

Notes:
Area $\qquad$

| Result 8 |  |
| :--- | :--- |

## 9. FLAMES SURROUNDING COOKPOT

## Equipment: Cookpot

## Procedure:

a) Keep cookstove fully ablaze from previous tests.
b) Place cook pot into cooking position.
c) Observe the amount of uncovered flames surrounding the cookpot and record a description.
d) Compare description with table to find rating.
e) Remove cook pot.

| Rating | Amount of Uncovered Flames Touching Cookpot |
| :---: | :---: |
| Poor | entire cook pot and/or handles |
| Fair | most of cook pot, not handles |
| Good | less than 4 cm up the sides, not handles |
| Best | none |

## Description

$\qquad$


Notes:
10. FLAMES EXITING FUEL CHAMBER, CANISTER, OR PIPES

Equipment: None

## Procedure:

a) Keep cookstove fully ablaze from previous tests. b) Visually inspect the amount, if any, of flames coming out of the fuel chamber, canister or pipes and record ifflames do or do not protrude. c) Consult table to find rating.

| Rating | Occurrence of Fire |
| :---: | :---: |
| Poor | Flames protrude |
| Best | Flames are contained |



## Description

$\qquad$

Notes:

## Overall Cookstove Safety Rating

To calculate the overall cookstove safety rating, place the point value of each individual rating in the "Value" column. Next multiply the individual ratings by their respective weights and place result in "Total" column. Sum these values and place that number in the box SUM. This value is applied to the overall rating metric to provide the overall safety rating of the stove.


SHell FOUNDATION


[^0]:    Hobo Data Logger: 2 at $3 \mathrm{ft}(1 \mathrm{~m}), 3$ at $4.5 \mathrm{ft}(1.4 \mathrm{~m}), 1$ at $7.5 \mathrm{ft}(2.3 \mathrm{~m})$ $\mathrm{CO}_{2}$ meter at $4.5 \mathrm{ft}(1.4 \mathrm{~m})$, All $4.3 \mathrm{ft}(1.3 \mathrm{~m})$ horizontally from stove Particulate meter at $3 \mathrm{ft}(1 \mathrm{~m})$

[^1]:    ${ }^{5}$ Aprovecho/GTZ. (1984). Charcoal: small scale production and use.
    ${ }^{6}$ Baldwin, S./VITA. Biomass stoves: engineering design, development and dissemination. Princeton University, 1986. P13.

[^2]:    ${ }^{7}$ Dr Ballard-Tremeer graduate thesis. (1997). See http://ecoharmony.com/thesis/PhDintro.htm.
    ${ }^{8}$ REPP Stove List, May 2002
    ${ }^{9}$ Dr. Winiarski Capturing Heat I (Aprovecho Research Center, 1996).

[^3]:    ${ }^{10}$ Nathan Johnson graduate thesis (Iowa State University 2005). See http://www.vrac.iastate.edu/~atlas/safety.htm.

